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Preparation and characterization of a novel antimicrobial film dressing for wound healing application

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Antibacterial activity and good mechanical properties are some of the characteristics required for an appropriate film dressing. A novel polymer blend was developed for wound healing application. Twenty-four formulations using the polymers chitosan, poly(vinyl alcohol) and/or ε -Polylysine and the plasticizer glycerol were designed using factorial design and then the films were prepared by the casting/solvent evaporation method. Seventeen films were obtained among the twenty-four proposed formulations that were characterized by Field Emission Scanning Electron Microscopy (FE-SEM) and Fourier Transform Infrared Spectroscopy (FTIR). Mechanical properties, such as tensile strength (σ), elongation at break (ε) and Young's modulus (Y) as well as antibacterial properties were determined. The best candidate was then further analyzed with regard to porosity, Water Vapor Transmission Rate (WVTR), swelling and cytotoxicity experiments. The results showed a film with semi-occlusive characteristics, good mechanical properties and no toxic. Incorporation of ε -Polylysine increased antibacterial activity against gram-negative (*Escherichia coli*) and gram-positive (*Staphylococcus aureus*) bacteria.

Keywords: Chitosan. Poly(vinyl alcohol). ɛ-Polylysine. Semi-occlusive dressing.

INTRODUCTION

Skin is a tissue that covers the body and it is the main barrier against external threats e.g. pathogens and mechanical shocks (Zhang Q *et al.*, 2019; Morgado, Aguiar-Ricardo, Correia, 2015; Radner, Fischer, 2014). Loss of skin integrity can occur due to traumas, burns, surgical procedures and others (Shahzad *et al.*, 2015; Antunes *et al.*, 2015). Wound healing is a complex biological process that aims to reestablish the skin integrity replacing the lost tissue (Ribeiro *et al.*, 2013;

Rodrigues et al., 2012; Cardoso et al., 2011). Some wounds have a self-healing capacity, but others need additional assistance for regeneration. Also, the healing process can be compromised by infections (Antunes et al., 2015; Kim et al., 2015; De Cicco et al., 2014). Therefore, after damage, skin must be covered by a wound dressing that act as barrier to prevent microorganism entry and fluid loss and to allow cell growth and gaseous exchanges (Shi et al., 2018; Li et al., 2017; Ribeiro et al., 2013). An ideal dressing must promote healing, be biocompatible, biodegradable and with antibacterial properties in addition to have suitable flexibility to cover the wound and to maximize patient comfort (Morgado, Aguiar-Ricardo, Correia, 2015; Vowden, Vowden, 2014; Ribeiro et al., 2013; Abdelrahman, Newton, 2011; Sung et al., 2010).

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Chitosan is a natural polymer that has been extensively used for dressing production. This polymer shows desirable characteristics for a dressing such as promotion of wound healing, biocompatibility and biodegradability (Avcu *et al.*, 2019; Dellera *et al.*, 2014; Hermans *et al.*, 2014; Lin *et al.*, 2013; Hosseini *et al.*, 2013; Giovino *et al.*, 2012; Kim *et al.*, 2011). Chitosan also has the ability to inhibit bacterial growth (Unnithan *et al.*, 2014). This particularly property results from chitosan's positively charged amino groups that interact with negatively charged groups present on bacterial surface (Kong *et al.*, 2010).

However, natural polymers show low mechanical strength. This issue can be solved by incorporating a synthetic polymer into the dressing. Poly(vinyl alcohol) (PVA) is a biodegradable and biocompatible synthetic copolymer widely used in dressing production (Ghalei, Asadi, Ghalei, 2018; Mogoşanu, Grumezescu, 2014; Kavoosi et al., 2014; Maleki, Gharehaghaji, Dijkstra, 2013; Cozzolino et al., 2012; Sung et al., 2010). The antibacterial property of chitosan can be improved by increasing the positive charges in its chain (Xiao et al., 2011). In this context, *ε*-Polylysine is a positively charged polyamine acid that shows broad spectrum antimicrobial activity (Zhang ZH et al., 2019). It is a biodegradable and non-toxic polymer that exhibits good water solubility and stability (Lopez-Pena, McClements, 2014; Zhang et al., 2012; Chang et al., 2010). In order to enhance mechanical properties, glycerol can be used in dressing formulations as a plasticizer (Ma et al., 2017; Ifuku et al., 2014; Fundo et al., 2014; Kammoun et al., 2013).

In the present study, film dressings using chitosan, glycerol, PVA and/or ϵ -Polylysine were prepared and characterized for wound healing applications.

MATERIAL AND METHODS

Material

Chitosan (medium molecular weight, ~75% deacetylated), PVA (molecular weight of 13 to 23 kDa, 89% hydrolyzed), potassium bromide and

3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2Htetrazolium bromide (MTT) were purchased from Sigma-Aldrich (São Paulo, Brazil). ε-Polylysine was purchased from Zhengzhou Sigma Chemical (Zhengzhou, China). Glycerol and glacial acetic acid were purchased from Biotec (São Paulo, Brazil). RPMI 1640 medium and Fetal Bovine Serum (FBS) were purchased from Vitrocell (Campinas, Brazil). Fibroblasts 3T6-Swiss Albino were obtained from Rio de Janeiro Cell Bank. *Escherichia coli* (NEWP 0022) and *Staphylococcus aureus* (NEWP 0023) were purchased from Newprov (Curitiba, Brazil). All reagents were used without further purification.

Factorial Design

In order to evaluate the mechanical properties (dependent variables), the experiments were performed using three 2^3 factorial designs. Three independent variables and their influence on the films were evaluated. Each independent variable was set at two levels. The lower and upper levels are represented by (-) and (+), respectively (Table I). The three independent variables in the first factorial design were chitosan, PVA, and glycerol. In the second, the three independent variables were chitosan, ϵ -Polylysine, and glycerol. The three independent variables in the third factorial design were chitosan, PVA, and ϵ -Polylysine. In this case, the ratio of glycerol that best developed the previous films was fixed (Table II).

TABLE I - Values for the lower level (-) and upper level (+) of independent variables evaluated in the factorial designs

Independent variable	(-) (mL)	(+) (mL)
Chitosan solution 1.5% w/v	10	30
PVA solution 3% w/v	10	30
ϵ -Polylysine solution 3% w/v	10	30
Glycerol	0.5	1

PVA: poly(vinyl alcohol)

Formulation	Chitosan (1.5%	5 w/v)	PVA (3% w/v)	ε-Polylysine (3% w/v)	Glycerol
FD01	10 mL	10 mL	-	0.5 mL	
FD02	30 mL	10 mL	-	0.5 mL	
FD03	10 mL	30 mL	-	0.5 mL	
FD04	30 mL	30 mL	-	0.5 mL	
FD05	10 mL	10 mL	-	1 mL	
FD06	30 mL	10 mL	-	1 mL	
FD07	10 mL	30 mL	-	1 mL	
FD08	30 mL	30 mL	-	1 mL	
FD09	10 mL	-	10 mL	0.5 mL	
FD10	30 mL	-	10 mL	0.5 mL	
FD11	10 mL	-	30 mL	0.5 mL	
FD12	30 mL	-	30 mL	0.5 mL	
FD13	10 mL	-	10 mL	1 mL	
FD14	30 mL	-	10 mL	1 mL	
FD15	10 mL	-	30 mL	1 mL	
FD16	30 mL	-	30 mL	1 mL	
FD17	10 mL	10 mL	10 mL	1 mL	
FD18	30 mL	10 mL	10 mL	1 mL	
FD19	10 mL	30 mL	10 mL	1 mL	
FD20	30 mL	30 mL	10 mL	1 mL	
FD21	10 mL	10 mL	30 mL	1 mL	
FD22	30 mL	10 mL	30 mL	1 mL	
FD23	10 mL	30 mL	30 mL	1 mL	
FD24	30 mL	30 mL	30 mL	1 mL	

TABLE II – Three 2³ factorial designs to films using chitosan, glycerol, PVA and/or ε-Polylysine

PVA: poly(vinyl alcohol)

Preparation of the Films

Using the factorial designs, twenty-four films were prepared by casting/solvent evaporation method (Murguía-Flores *et al.*, 2016; Tan *et al.*, 2008; Azad *et al.*, 2004). Chitosan solution (1.5% w/v) was prepared by dissolving it in 0.5% (v/v) aqueous glacial acetic acid. Solutions of PVA (3% w/v) and ε -Polylysine (3% w/v) were prepared by their dissolution in deionized water. According to the Table II, chitosan, PVA and ε -Polylysine solutions and glycerol were mixed under magnetic stirring, at 45 °C and 3 h. After that, equal volumes of each formulation were transferred into polyethylene Petri dishes and dried at 37 °C, until constant weight.

Characterization

Field Emission Scanning Electron Microscopy

Surface of the obtained films and thickness of the selected film (with best mechanical and antimicrobial properties) were performed on Field Emission Scanning Electron Microscopy (FE-SEM) (Mira3 - TESCAN[®]) at 5-10 kV.

Fourier Transform Infrared Spectroscopy

Chemical characterization of the obtained films was done by Fourier Transform Infrared Spectroscopy (FTIR) (IR Prestige 21 - SHIMADZU[®]) in the range of 4000-400 cm⁻¹. Potassium bromide pallet method was applied.

Mechanical Properties Analysis

Mechanical properties of obtained films were determined on Universal Mechanical Testing Machine (AG-I 10kN - SHIMADZU[®]). Films samples (40 mm x 15 mm) were held between clamps and pulled at a rate of 1.0 mm.min⁻¹. Tensile strength (σ), elongation at break (ε) and Young's modulus (Y) were determined (Lin *et al.*, 2013). The experiments were performed in triplicate. Statistical analysis of the data was performed using ANOVA and Tukey test at a significance level of p<0.05 and the results were presented as mean \pm standard deviation.

Antibacterial Activity Analysis

Antibacterial activity of obtained films was assessed by agar diffusion method (Patil *et al.*, 2016). Suspensions of *Escherichia coli* and *Staphylococcus aureus* bacteria in a final concentration of 10⁸ CFU. mL⁻¹ (MacFarland scale) were streaked evenly in three planes, using a cotton swab, on the surface of Petri dishes containing Mueller-Hinton agar. Films samples (6 mm diameter disks) were placed into Petri dishes that were incubated overnight at 37 °C. After incubation, antibacterial activities of the films were determined by measuring the inhibition zones formed. The experiments were performed in triplicate.

Porosity Test

The porosity ratio (P%) of the selected film, was measured through the determination of the amount of solvent absorbed by the film sample (20 mm x 20 mm) after 1 h of immersion in ethanol, using Eq. 1 (Correia *et al.*, 2013):

$$P\% = \frac{Ww - Wd}{d_{ethanol}V_{film}} x \ 100 \tag{1}$$

where, W_w and W_d are the weight of the wet and dry film, respectively, $d_{ethanol}$ is the density of the ethanol at room temperature and V_{film} is the volume of the wet film. V_{film} was determined by measuring the film thickness with a pachymeter. The experiment was performed in triplicate.

Water Vapor Transmission Rate Test

Water Vapor Transmission Rate (WVTR) measurement of the selected film was determined using a usual method (Antunes *et al.*, 2015; Lin *et al.*, 2013). Briefly, film samples were used to seal the opening of a glass test tube (1.77 cm²) containing 10 mL of deionized water. Tape was used to attach the film. The tubes were left in an incubator at 37 °C. After 24 h, water evaporation was determined by weight measurement. WVTR was calculated using Eq. 2:

$$WVTR = \frac{W_{loss}}{A} \tag{2}$$

where, W_{loss} is the weight loss of water and A is the area of the tube opening. Non-attached films were

used as controls. The experiment was performed in quintuplicate.

Swelling Test

The water uptake was determined for the selected film by the following method (Hermans *et al.*, 2014; Sung *et al.*, 2010): film samples (20 mm x 20 mm) were weighed. The samples were left in a beaker containing 20 mL of phosphate buffered saline (PBS) pH 7.4 at 37 °C for 4 h. After this period, excess amount of PBS was removed and the film samples wet were weighed. Swelling ratio (S%) was calculated according Eq. 3:

$$S\% = \frac{Ww - Wd}{Wd} \times 100 \tag{3}$$

where, W_w and W_d are the weights of the wet and dry film, respectively. The experiment was performed in triplicate.

Cytotoxicity Analysis

Cytotoxicity evaluation of the materials used for preparation of selected film was performed *in vitro* using MTT assay (Mosmann, 1983). Briefly, Fibroblasts 3T6-Swiss Albino in RPMI 1640, supplemented with 10% FBS, were seeded into 96 well culture plates (2.5 x 10⁵ cell/well) followed by incubation at 37 °C and 5% CO₂ during 24 h for adhesion. Then, was added RPMI 1640, supplemented with 10% FBS, containing the same amounts of chitosan, ϵ -Polylysine and glycerol used for selected film preparation, followed by incubation at 37 °C and 5% CO₂. After 24 h, MTT solution (0.5 mg.ml⁻¹) in RPMI 1640, supplemented with 10% FBS, was added and cells were incubated for 1 h at 37 °C and 5% CO₂. Subsequently, DMSO was added to solubilize the formazan crystals formed and the absorbance was measured at 550 nm using a plate reader (μ QuantTM -BIOTEK[®]). Statistical analysis of the data was performed using ANOVA and Tukey test at a significance level of p<0.05.

RESULTS AND DISCUSSION

Eight films were prepared to each factorial design. Seventeen films were obtained among the twenty-four proposed formulations (FD01, FD02, FD03, FD04, FD05, FD06, FD07, FD08, FD09, FD10, FD14, FD16, FD17, FD18, FD20, FD22, FD24). Visually, the films presented smooth surfaces, indicating homogeneity after the combination of polymers and plasticizer. All films showed a transparent appearance and were flexible.

It was observed at the images obtained by FE-SEM (Figure 1) that some formulations showed smooth surfaces (FD02, FD03, FD04, FD07, and FD08), some showed rough surfaces (FD01, FD05, FD06, FD09, FD10, FD14, FD16, FD17, FD20 and FD22) and others showed glycerol excess that was not incorporated into the film (FD18 and FD24), which could also be observed macroscopically. All the obtained films presented small clusters (clear dots in the images), which are speculated to be composed of not soluble portions of polymers.



FIGURE 1 - FE-SEM surface micrographs of A) FD01; B) FD02; C) FD03; D) FD04; E) FD05; F) FD06; G) FD07; H) FD08; I) FD09; J) FD10; K) FD14; L) FD16; M) FD17; N) FD18; O) FD20; P) FD22; Q) FD24.

FTIR was used to evaluate the chemical characteristics of the novel films. Figure 2 shows the FTIR spectra of chitosan, ε-Polylysine, PVA and some obtained films, as examples. The large band observed around 3351 cm⁻¹ indicates chitosan amino and hydroxyl groups. FTIR spectrum of PVA reveals a major peak at 2858 cm⁻¹ which is associated with C-H broad alkyl stretching and the typical strong hydroxyl band at 3450 cm⁻¹ for free alcohol. Amide group is a main functional group of *ɛ*-Polylysine. It was observed in *ɛ*-Polylysine spectrum characteristic amide bands at 1660 cm⁻¹, mainly CO group stretching, and at 1564 cm⁻¹ related to NH group. The analysis of each film spectra reveals the characteristic peaks of all the components used in their preparation. The FTIR spectra obtained for the films showed bands related to the same groups observed for their initial materials, which suggest that there was not formation of new covalent bonds between the polymers and the starting materials are compatible to each other.



FIGURE 2 - FTIR spectra of chitosan, ε-Polylysine, PVA and some obtained films (FD05, FD06, FD09, FD10, FD18, FD22).

In order to investigate the mechanical properties of obtained films, tensile strength (σ), elongation at break (ε) and Young's modulus (Y) (dependent variables of factorial design) were determined and are summarized in Table III. Tensile tests are performed to evaluate the maximum stress sustained by a film (Kavoosi *et al.*, 2014). For wound treatment, high tensile strength values are required to maintain film integrity and high elongation at break values show flexibility in order to make the application easy on skin (Morgado, Aguiar-Ricardo, Correia, 2015). Young's modulus measure the relationship between the stress (applied force) and strain (resulting deformation) of the films.

TABLE III - Mechanical properties of obtained films

Test	σ (N/mm ² ± SD*)	ε (mm± SD)	Y (N/mm ² ± SD*)
FD01	1.06 ± 0.330	43.20 ± 15.456	2.44 ± 0.520
FD02	5.46 ± 0.928	134.71 ± 11.207	3.66 ± 0.162
FD03	1.73 ± 0.284	37.96 ± 11.165	7.24 ± 0.119
FD04	1.78 ± 0.539	31.31 ± 7.595	7.03 ± 1.901
FD05	0.90 ± 0.182	27.01 ± 6.485	2.68 ± 0.558
FD06	1.01 ± 0.258	40.79 ± 15.432	2.67 ± 0.219
FD07	0.93 ± 0.229	24.19 ± 1.623	5.32 ± 1.582
FD08	1.18 ± 0.376	37.36 ± 16.935	3.49 ± 0.204
FD09	1.42 ± 0.057	60.72 ± 9.784	2.28 ± 0.230
FD10	3.38 ± 1.411	115.06 ± 33.150	2.30 ± 0.255
FD14	1.04 ± 0.135	67.75 ± 5.621	1.52 ± 0.097
FD16	0.54 ± 0.160	33.88 ± 9.880	1.53 ± 0.268
FD17	0.21 ± 0.038	20.63 ± 3.115	0.90 ± 0.167
FD18	1.21 ± 0.565	64.02 ± 32.091	2.03 ± 0.233
FD20	0.88 ± 0.0640	60.78 ± 8.280	1.77 ± 0.174
FD22	1.03 ± 0.388	45.45 ± 24.384	2.95 ± 0.446
FD24	0.71 ± 0.182	54.39 ± 23.89	1.70 ± 0.178

*Standard deviation calculated after three experiments

The results of the factorial design showed different values of σ and ε for the different film dressings prepared. However, in general, all films presented low values of Y which means that little stress cannot cause permanent deformation, implying a flexible material and suitable for the required destination. FD02 and FD10 showed the higher σ and ε values, and were significantly higher than the other films (Figure 3). Since the goal of the factorial design was to find the films with best's σ and ε values, FD02 and FD10 were considered the bests obtained films dressings.



FIGURE 3 - Comparison of A) tensile strength (N/mm²) and B) elongation at break (%) of obtained films.

Both FD02 and FD10 were obtained using chitosan upper level and glycerol lower level, which suggest that, in this ratio, glycerol produces the desired plasticizer effect. FD02 was prepared with lower level of PVA and FD10 with lower level of ε -Polylysine. Comparing these two films, the film prepared using PVA (FD02) showed better tensile strength and elongation at break characteristics, which confirm that PVA gives a greater improvement of the mechanical properties.

Other important property to be evaluated during the film dressing developed process is the antibacterial activity. Antimicrobial agents, such as silver sulfadiazine, have been added to film dressing formulations to offer this property, but since there are some concerns about the toxicity of these components, biocompatible agents are preferred (Morgado, Aguiar-Ricardo, Correia, 2015). Antibacterial activities of the obtained film dressings were assessed through agar diffusion method against *Staphylococcus aureus* and *Escherichia coli bacteria*. These two bacterial strains represent both gram-positive and gram-negative pathogens, respectively (Antunes *et al.*, 2015; Karami *et al.*, 2013). Formation of an inhibition halo (mm) was evaluated and is shown in Table IV.

TABLE IV - Inhibition zone diameter (mm) obtained against *S. aureus* and *E. coli* using the films prepared

Test	S. aureus	E. coli
FD01	6*	6
FD02	6	6
FD03	6	6
FD04	6	6
FD05	6	6
FD06	6	6
FD07	6	6
FD08	6	6
FD09	12	12
FD10	10	10
FD14	10	10
FD16	12	12
FD17	10	12
FD18	10	10
FD20	12	12
FD22	10	10
FD24	12	10

*6mm is the diameter of the disc

Between the two films that presented appropriate mechanical properties (FD02 and FD10), FD02, consisting of chitosan (upper level), PVA (lower level) and glycerol (lower level), showed no inhibition halo, but no bacteria colonized the film. Whereas FD10, consisting of chitosan (upper level), ɛ-Polylysine (lower level) and glycerol (lower level), showed an inhibition halo of 10 mm. Similar film dressing showed inhibition halos of 10.4 mm against S. aureus 7.8 mm against E. coli. (Karami et al., 2013). All the films containing ε-Polylysine presented the same results as FD10, which highlights the importance of *ɛ*-Polylysine in chitosan films for enhancing the bactericidal effect. Since the FD10 formulation showed good mechanical properties in addition to efficient antimicrobial properties, the film was further analysed.

A moist wound environment is essential to improve the healing process because it increases the reepithelialization velocity (Abdelrahman, Newton, 2011). Therefore, suitable thickness and porosity of film dressing are important to allow cellular infiltration and proliferation, as well as for gas and fluid exchange (Morgado, Aguiar-Ricardo, Correia, 2015; Antunes et al., 2015). Cross-sectional morphology and thickness (at five different regions) of FD10 were determined by FE-SEM (Figure 4) and the average thickness found was $82.72 \pm 21.92 \,\mu\text{m}$. The selected film presented a porosity ratio of 27.2% and the WVTF test result confirmed the low porosity ratio found in the porosity test result. These two tests are complementary since the larger is the porosity of the film, the greater is the loss of humidity of the wound environment (high WVTF value).

The WVTR values recommended to prevent excessive dehydration are between 2000-2500 mL/m^2

day (Antunes *et al.*, 2015; Sung *et al.*, 2010). However, some available commercial film dressings, like OpSites[®], present WVTR around 792 mL/m² day, which is close to the data obtained for FD10 (784.2 mL/m² day) (Figure 5). This result can be influenced by structural properties of the film dressing and external conditions such as humidity and temperature (Morgado, Aguiar-Ricardo, Correia, 2015).

This data was obtained using the average value after five experiments (0.1388 mL/day) in Eq. 2. This result, in addition to the measured porosity ratio, characterize the film as semi-occlusive (300-800 mL/m²/ day) (Vowden, Vowden, 2014; Abdelrahman, Newton, 2011). There is some resistance towards the use of semiocclusive dressings on wound treatment with specific concerns related to bacterial growth. However, several studies comparing infection rates between occlusive and non-occlusive dressings have shown better result for the occlusive ones (Vowden, Vowden, 2014). It seems that these dressings prevent bacterial ingress into the wound by producing an acidic environment that retards bacterial growth. In addition, semi-occlusive films trap moisture, creating better healing environment.

Swelling tests (water uptake) allows a quantitative evaluation of the exudates absorption ability of the films which also confirms if the dressing can maintain the moist environment of the wound without allowing the accumulation of exudate, which also impairs healing. An ideal film dressing must exhibit water uptake between 100-900% (Morgado, Aguiar-Ricardo, Correia, 2015; Lin *et al.*, 2013). FD10 presented a swelling ratio of 147%, which is within the suitable range for this application.



FIGURE 4 - Cross-sectional FE-SEM micrographs of FD10 in five different regions.



FIGURE 5 - Weight loss (mg) obtained for FD10.

For film dressing purposes, a material must be biocompatible, which can be evaluated through *in vitro* studies, using appropriate cell line models. *In vitro* cytotoxicity assay was performed to evaluate the toxicity of the initial materials used to prepare the selected wound dressing film FD10. This study was conducted using the MTT assay to determine the cell viability comparing to a control (culture medium). As shown in Figure 6, the cells showed high viability, maintaining their metabolic activity, after treatment with each initial material in the concentrations used for film preparation. There was no statistical difference between the samples and control. The cell survival was expected since the materials tested are known to be biocompatible.



FIGURE 6 - Cell viability of fibroblasts 3T6-Swiss Albino treated with chitosan, ε-Polylysine and glycerol (MTT assay).

CONCLUSION

In this study were prepared seventeen film dressings using chitosan, glycerol, PVA and/or ε -Polylysine after factorial designs. Lower levels of glycerol seem to be sufficient for its use as plasticizer. Two formulations showed good mechanical properties for film dressing application. Between these two, the film containing ε -Polylysine showed greater antibacterial activity, demonstrating a synergistic effect with chitosan against both gram-positive and negative bacteria. The selected film also exhibited good swelling properties and it was determined to be a semi-occlusive film. Additionally, the absence of toxicity makes the FD10 a novel film dressing potentially effective to wound healing aplication.

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