



Postharvest ascorbic acid application maintained physiological and antioxidant responses of Guava (*Psidium guajava* L.) at ambient storage

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Abstract

Guava (*Psidium guajava*) is a highly perishable fruit with a short shelf life as physico-chemical changes occur continuously and rapidly after harvest leading to heavy postharvest losses. This experiment was laid down to unravel the effects of ascorbic acid (AA) on ripened guava fruits to improve its shelf life and quality. Four different concentrations of AA namely 0, 50, 100, 200 ppm were used to treat guava fruit for five minutes followed by storage at 25±2 °C and ≥80% relative humidity (RH) for 12 days. Physiological and biochemical changes were studied, together with the specific enzymatic activities for catalase (CAT), superoxide dismutase (SOD) and peroxidase (POD). The results revealed that ascorbic acid treatments significantly reduced PWL, fruit decay percentage and suppressed pH and sugar/acid ratio than control fruits. Furthermore, soluble solid content, total acidity, total sugar, vitamin C and total phenolic contents were recorded higher in 200 ppm AA-treated fruits and maintained higher eating quality than control fruits. In addition, SOD, POD and CAT activities were observed higher in 200 ppm AA-treated fruits than control. The results clearly demonstrate that 200 ppm AA application have a potential to improve quality attributes of guava fruit.

Keywords: ascorbic acid; biochemical changes; enzyme activities; shelf life; guava.

Practical Application: Application of ascorbic acid improves the quality and shelf life of guava fruit during storage at ambient temperature.

1 Introduction

Guava (*Psidium guajava* L.) is valued fruit crop commercially grown in tropical and sub-tropical zones of world (Watson & Dallwitz, 1991). The significance of guava is due to its enchanting taste and remarkable nutritional properties as it has five times as much ascorbic acid as citrus fruit (McCook-Russell et al., 2012) and also contains different kinds of essential bioactive compounds (Gutiérrez et al., 2008). Guava has high levels of pectin, dietary fiber, vitamins, antioxidant and mineral contents as compared to the other fruits and, therefore, is used in medicine to cure gastroenteritis, dysentery, healing of wounds and ulcers rheumatics (Olajide et al., 1999).

Guava is climacteric fruit with steep respiration peak and high rate of ethylene production that limits its postharvest shelf-life to three to four days at room temperature whereas, on the other hand refrigerated storage causes chilling injury (Murmu & Mishra, 2017). Therefore, guava export is limited

in the world due to its rapid susceptibility to damages and low shelf life. It has been reported that postharvest losses in guava are higher than pre-harvest losses (Gill et al., 2016). Membrane degradation and browning symptoms are induced by accumulation of reactive oxygen species (ROS) during stress conditions (Mahajan et al., 2017). Previously, various strategies were used to prolong the shelf life of guava such as treatment with edible coatings (Silva et al., 2018), preharvest application of aqueous hexanal (Gill et al., 2016), gamma-irradiation (Pandey et al., 2010), calcium salts (Javed et al., 2016), 1-MCP (Phebe & Ong, 2010; Xing et al., 2010), control atmosphere storage (Teixeira et al., 2016), low temperature storage (Mahajan et al., 2017) and packaging types (Rana & Siddiqui, 2018).

Ascorbic acid (AA) is the most abundant antioxidant in nature. AA and its derivatives have been used as an antioxidant and anti-browning agent in edible coatings to retain postharvest quality

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of fresh-cut fruits and vegetables (Tapia et al., 2008; Xing et al., 2010). AA in combination with calcium salts and organic acids prevent browning and membrane breakdown by controlling the activity of polyphenol oxidase (Oms-Oliu et al., 2010). AA has also shown antibacterial properties for fresh-cut banana (Yurdugül, 2016), apple (Qi et al., 2011) and papaya (Tapia et al., 2008). Mostly, AA and its derivatives have been used as anti-oxidative, anti-browning and antibacterial agent (Sogvar et al., 2016). However, effect of AA on postharvest quality attributes and responses of antioxidant system of guava during ambient storage have not been explored yet. Therefore, this experiment was proposed with the aim to reveal how ascorbic acid effect on physicochemical changes, quality attributes, and antioxidative enzymatic activity in guava cultivar "Golla" during ambient storage.

2 Materials and methods

2.1 Collection Fruit Sample

Guava fruits (cv. Golla) were freshly harvested from Postgraduate Agriculture Research Station (PARS), Faisalabad, during the month of November and promptly shifted to the laboratory for analysis. Selected fruits were uniform in size, color and maturity stage. Healthy fruits were sorted based on visible symptoms of defect, decay or disease. The solution was prepared by dissolving ascorbic acid (99%, Sigma-Aldrich, USA) in deionized water. Four different concentrations of ascorbic acid were prepared viz. 0, 50, 100, 200 ppm. The selected fruits were washed, dried and dipped in respective ascorbic acid solutions for five minutes. Fruits were air dried before storing at ambient temperature (25 ± 2 °C) and $\geq 80\%$ RH. The experimental design was completely randomized design along with factorial arrangements and consisted of four treatments (i.e. 0, 50, 100, 200 ppm AA) with three replications. Each treatment was comprised of 40 fruits per replication. Fruit samples, each comprising of 9 fruits, were drawn from each treatment at 3, 6, 9 and 12 days of storage for physicochemical analysis. Following observations were made to find out the impact of AA application on the quality of stored guava fruits.

2.2 Determination of Physiological and Biochemicals Variables

Physiological weight loss (PWL) was calculated by weighing 10 fruits per treatment on digital balance (PTL, RX 5000, Japan) before and after storage (Waskar et al., 1999). PWL was calculated according to the formula as given below.

$$\text{Physiological weight loss (\%)} = \frac{\text{Weight before storage} - \text{weight after storage}}{\text{Weight before storage}} \times 100.$$

Similarly, fruit decay percentage was estimated by taking the ratio between number of decayed fruit and total number of fruit.

An advanced refractometer (ATAGO, RS-5000, Atago, Japan) was utilized to gauge soluble solid content (SSC) of fruit juice. The TA of fruit juice was analyzed as suggested by (Hortwitz, 1960). Distilled water and 0.1 N NaOH was used for dilution of fruit juice for titration, using 2-3 drops of phenolphthalein as an

indicator. To calculate the SSC: TA ratio the percentage of SSC was divided with the percentage of the TA.

Digital pH meter was used to determine the pH (HI 98107, Hanna Instruments, Mauritius). Vitamin C contents were measured from guava juice following the previously described method by (Hortwitz, 1975). Fruit juice was filtered and titrated using 2, 6-dichlorophenolindophenol dye to light pink color as end point following the dilution with 0.4% oxalic acid solution. For sugar analysis, guava fruit pulp (10 g) was added to 100 mL distilled water (25 mL 25% lead acetate solution and 10 mL 20% potassium oxalate) followed by filtration of solution. The filtrate was used to determine total sugars and were expressed as percentage. Measurement of total phenolic contents from guava pulp were quantified by the method proposed by (Ainsworth & Gillespie, 2007). Gallic acid was used as standard and their concentration was expressed on fresh weight basis mg Kg^{-1} . Standard used in this study was gallic acid its concentration was presented on fresh weight basis.

2.3 Determination of SOD, POD and CAT Enzymes activities

Fruit pulp (10 g) was homogenized in 25 mL of ice-cold extraction buffer and 0.5 g polyvinyl polypyrrolidone (PVPP). Extraction buffer for SOD, CAT assays (50 mM sodium phosphate with pH 7.8) and for POD (100 mM sodium phosphate buffer with pH 6.4) were used. The homogenate was centrifuged ($27,000 \times g$ for 50 min at 4 °C) and the resulting supernatants were used directly for assay. SOD enzyme was determined as described by (Liu et al., 2014). Briefly, 3 mL reaction mixture (65 mM sodium phosphate buffer (pH 7.8), 13 mM methionine, 75 μM NBT, 10 μM EDTA, 2 μM riboflavin) and 0.1 mL of the enzyme extract were used for SOD activity. The mixtures were illuminated to light ($60 \text{ mol m}^{-2} \text{ s}^{-1}$) for 10 min and the absorbance was noted at 560 nm. Identical solution held in the dark served as blank. Finally, SOD activity was expressed $\text{Ug}^{-1} \text{ min}^{-1} \text{ FW}$. POD enzyme was determined as described by (Ali et al., 2016). Briefly, a reaction mixture (100 mM sodium phosphate buffer with pH 6.4), 8 mM guaiacol) was prepared and 100 μL enzyme extract was incubated at 30 °C and added with H_2O_2 . Finally, the absorbance was determined at 460 nm, and enzyme activity was expressed as $\text{Ug}^{-1} \text{ min}^{-1} \text{ FW}$.

CAT enzyme activity was determined as method proposed recently by (Liu et al., 2014). Briefly, 1.9 mL reaction solution (50 mM sodium phosphate buffer with pH 7, 1 mL 40 mM H_2O_2) and 0.1 mL enzyme extract was used for CAT activity. About 200 μL from the above solution was placed in the 96 well plate and absorbance was taken at 240 nm and enzyme activity was expressed as $\text{Ug}^{-1} \text{ min}^{-1} \text{ FW}$.

2.4 Statistical analysis

Data was subjected to analysis of variance using two factorial completely randomized design (CRD) with Statistix -8.1® software (Analytical Software, Tallahassee, USA). Comparison of Least significant differences among treatments mean was executed by using Fishers test ($P < 0.05$).

3 Results and discussion

The results of physiological and chemical analysis of guava fruits (cv. Golla) after harvest during storage at ambient conditions are shown below. Physiological weight loss (PWL) of guava fruits increased in all treatments with the progression of storage periods. PWL exhibited linear increasing trend during entire 12 days of storage (Figure 1A). Application of AA significantly ($p \leq 0.05$) retarded the PWL of guava fruits during storage compared to untreated fruits. The lowest PWL was found in 200 ppm AA-treated fruits, while highest PWL was observed in untreated fruits after 12 days of storage (Figure 1A). AA and its derivatives have been used as an antioxidant and anti-browning agent in edible coatings to retain postharvest quality of fresh-cut fruits and vegetables (Tapia et al., 2008; Xing et al., 2010). The results revealed that PWL increased during storage days, which might be due to increase in respiration rate from fruits. PWL decreased significantly all treatments throughout the storage periods, likewise, results have been recorded in apple fruits treated with 1% chitosan + 2% AA showed decline in weight loss than control fruits (Qi et al., 2011).

Fruit decay (FD) continuously increased in untreated fruits as compared to AA-treated guava fruits with progress of storage conditions (Figure 1B). AA-treated fruits showed significantly ($P \leq 0.05$) less fruit decay as compared with untreated fruits. However, decay percentage of untreated fruits was 2.27 times higher, than fruits treated with 200 ppm AA after 12 d of storage (Figure 1B). Our findings are similar to that found by Puthmee et al., (2009), which obtained postharvest treatment of fresh-cut mangoes with 1.5% AA reduces weight loss, change in color and microbial decay.

Regardless of the treatments, SSC of guava fruits increased slowly during storage periods (Table 1). SSC in AA treated samples were lower than untreated guava fruits during storage periods. SSC was increased gradually in all treatments, however, higher level was noticed in untreated fruits (9.96%), while lower level was found in 200 ppm AA treated fruits (8.27%) during the 12 d of storage (Table 1). Silva et al. (2018) obtained approximately $8 \pm 1.5\%$ SSC in moderately ripe or ripe guava fruits which also showed gradual increase during storage. AA treated fruits

significantly reduced the SSC contents than control. Similarly, SSC content were recorded lower in 300 ppm AA treated fruits of 'Umran' ber at room temperature (Siddiqui & Gupta, 1995). AA+ Aloe vera combination stabilized SSC content during first 12 days of storage and thereafter cause a slight increase in subsequent storage days (Sogvar et al., 2016).

The changes in TA contents were shown in the Table 1. TA contents continuously reduced regardless of treatments during entire storage time. However, TA contents found higher in AA treated samples as compare with control during all storage time. TA in 200 ppm AA treated fruit was higher (0.47%) than untreated fruits (0.32%) as shown in Table 1. According to Echeverria & Valich, (1989) alteration in fruit metabolism results in depletion of organic acids during respiration caused TA contents to decrease during storage. Arowora et al. (2013) found that ascorbic acid application might inhibit the uptake of oxygen during fruit metabolic process, which as result reduced the respiration rate. Sogvar et al. (2016) also observed that decline in TA was lower in AA treated fruits, while recorded higher in untreated fruits of guava, plum and ber

All treatments showed a gradual increasing trend for SSC: TA ratio with the increase in storage days (Table 1). However, in control ratio of SSC: TA was significantly ($P < 0.05$) higher, than all AA treatments during the all the storage interval. On 12 d of storage, 200 ppm AA showed significantly lowest SSC: TA ratio (1.34-times) than control fruit as shown in Table 1. Increasing the AA levels led to decrease in SSC: TA ratio. Similar finding has been reported by Ali et al., (2016), which showed significantly lower SSC: TA ratio in treated litchi fruits than control during cold storage.

There was slight changes in pH value for AA-treated and untreated fruits during the 12 d of storage are shown in Table 1. Regardless of treatments, pH values slightly increased during storage periods. On an average, the data indicated that pH was registered higher in untreated fruits and lower for AA-treatments with the progression of storage period (Table 1). Phebe & Ong (2010) found that low pH in fruits indicated in 1-MCP treated guava fruits which delayed the senescence than control fruits.

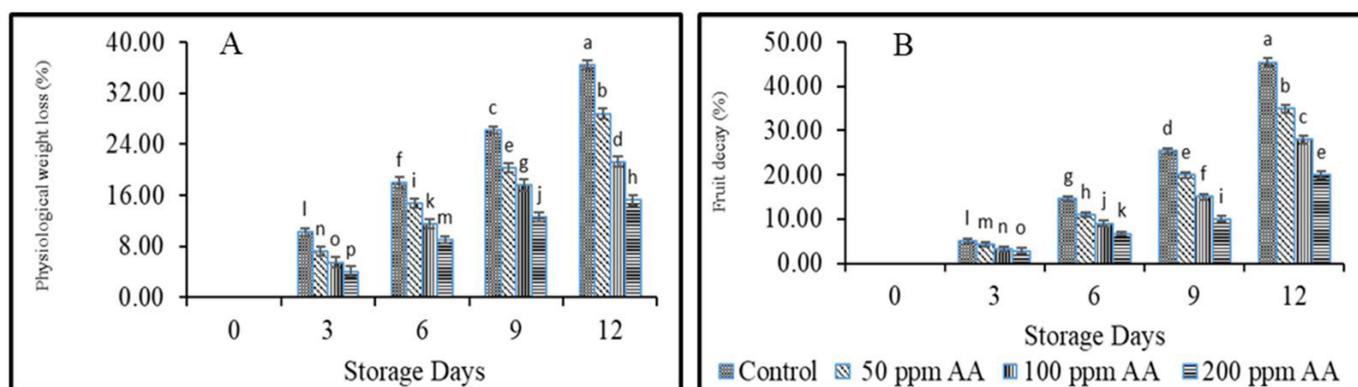


Figure 1. Effect of postharvest application of AA on physiological weight loss (A) and fruit decay (B) of guava fruit during storage at 25 °C for 12 days. Mean values in graph with Different letter are significantly different by Fisher's least significant (LSD) test ($P \leq 0.05$). Vertical bars indicate standard error, and each value is the mean of three replicates.

Table 1. Effect of postharvest application of AA on soluble sugar content (SSC), total acidity (TA), sugar/acid ratio (SSC: TA) and pH of guava fruit during storage at 25 °C for 12 days.

Attributes	Ascorbic acid (ppm)	Storage days				
		0	3	6	9	12
SSC (%)	0	7.3 ± 0.15 o	8.15 ± 0.18 k	8.99 ± 0.16 c	9.21 ± 0.13 b	9.69 ± 0.16 a
	50	7.3 ± 0.16 o	7.72 ± 0.19 l	8.50 ± 0.18 f	8.71 ± 0.17 e	9.08 ± 0.18 b
	100	7.3 ± 0.15 o	7.52 ± 0.14 m	8.35 ± 0.11 h	8.47 ± 0.17 g	8.90 ± 0.21 d
	200	7.3 ± 0.15 o	7.37 ± 0.20 n	8.05 ± 0.19 j	8.18 ± 0.20 i	8.27 ± 0.19 g
TA (%)	0	0.71 ± 0.004 a	0.58 ± 0.002 d	0.51 ± 0.004 g	0.45 ± 0.003 a	0.32 ± 0.002 m
	50	0.71 ± 0.003 a	0.61 ± 0.003 c	0.54 ± 0.003 h	0.51 ± 0.002 b	0.37 ± 0.003 l
	100	0.71 ± 0.004 a	0.62 ± 0.002 c	0.57 ± 0.003 e	0.53 ± 0.003 c	0.40 ± 0.004 k
	200	0.71 ± 0.004 a	0.68 ± 0.003 b	0.62 ± 0.004 c	0.57 ± 0.004 f	0.46 ± 0.005 j
SSC: TA (ratio)	0	10.28 ± 0.15 n	13.81 ± 0.16 j	17.52 ± 0.17 f	20.24 ± 0.21 d	30.00 ± 0.18 a
	50	10.28 ± 0.17 n	12.65 ± 0.20 k	15.74 ± 0.16 h	17.07 ± 0.22 g	24.54 ± 0.21 b
	100	10.28 ± 0.20 n	12.12 ± 0.19 l	14.64 ± 0.19 hi	15.68 ± 0.19 gh	22.25 ± 0.22 c
	200	10.28 ± 0.19 n	11.38 ± 0.17 m	13.66 ± 0.16 j	14.82 ± 0.16 i	19.58 ± 0.19 e
pH	0	2.69 ± 0.010 q	2.88 ± 0.010 m	3.39 ± 0.020 i	3.71 ± 0.030 a	3.65 ± 0.040 c
	50	2.69 ± 0.020 q	2.86 ± 0.030 n	3.32 ± 0.010 j	3.68 ± 0.020 b	3.51 ± 0.030 f
	100	2.69 ± 0.030 q	2.85 ± 0.020 o	3.12 ± 0.040 k	3.63 ± 0.010 d	3.47 ± 0.020 g
	200	2.69 ± 0.020 q	2.81 ± 0.040 p	3.01 ± 0.030 l	3.53 ± 0.030 e	3.41 ± 0.030 h

Different lettering within column shows significant differentness in mean values by Fisher's least significant (LSD) test ($P < 0.05\%$). Each value is the mean of three replicates

Kumar et al. (2012) observed higher pH due to decline in acidity in fruit juice, and guava has low pH due to high in organic acids.

The changes in total sugar (TS) for AA-treated and control during storage are shown in Table 2. TS were found increased in AA-treated fruits and control fruits up to 9 d of storage, thereafter declined slowly. However, TS contents was greater (6.33%) in 200 ppm AA treated fruits than untreated fruits (5.75%) on 12 d of storage. (Table 2). Our results are consistent with the previous findings that different postharvest chemicals (calcium compounds, GA_3) increased TS in guava fruits as compared with control during storage periods (Mahajan et al., 2017). It has been reported that increase in TS attributed to the fast disintegration of starch into sugar causing no further increase in TS indicating that organic serve as substrate during respiration (Javed et al., 2016; Mahajan et al., 2017; Wills et al., 1980).

In all treatments vitamin C gradually declines during the storage (Table 2). Nevertheless, the rate of decline in vitamin C were significantly higher in control as compared to AA treatments. On 12 d of storage, 200 ppm AA-treated fruits (123.34 mg/100g) showed higher vitamin C than control (101.88 mg/100g) as shown in Table 2. The reduction in vitamin C loss might be due to low oxygen availability that delays the harmful oxidation reaction of AA in food products (Tapia et al., 2008). Our results are similar with Xing et al., (2010), which reports increase in biosynthesis

or reduction in breakdown of vitamin C might have resulted in low rate of decline in AA treated fruits under storage.

TPC gradually decreased regardless of treatments during entire course of storage period. AA-treated fruits showed higher level of TPC, relative to control, at all sampling times. TPC contents found approximately 1.5 fold higher in 200 ppm AA-treated fruit (75.23 mg/kg), as compared with control (45.33 mg/kg) on 12 d of storage (Table 2). Singh & Pal (2008) observed that ascorbic acid treated guava fruits have higher TPC contents during storage. Vishwasrao & Ananthanarayan (2016) also obtained that TPC were found higher in HPMC-based edible coated fruits than uncoated guava fruits TPC were found higher at the harvest time then decline to lower limits during storage of guava fruits (Mahajan et al., 2017).

3.1 Effect of AA on SOD, POD and CAT Enzyme Activities

Regardless of the treatments, SOD activity increased first, then linearly decreased in guava fruits during storage periods. AA-treated fruits showed relatively higher SOD activity than untreated fruit during entire storage durations (Figure 2A). SOD activities significantly ($P \leq 0.05\%$) rose slowly as the AA levels increased than control. At 12 d of storage, SOD activity was 1.4-fold higher in 200 ppm AA-treated fruits, compared with control. Likewise, compared with control, 100 ppm AA and 50 ppm

Table 2. Effect of postharvest application of AA on total sugars (TS), Vitamin C contents (Vit. C) and total phenolic contents (TPC) of guava fruit during storage at 25 °C for 12 days.

Attributes	Ascorbic acid (ppm)	Storage days				
		0	3	6	9	12
TS (%)	0	5.70 ± 0.19 n	5.80 ± 0.20 l	5.95 ± 0.18 h	5.85 ± 0.17 k	5.75 ± 0.14 m
	50	5.70 ± 0.19 n	5.85 ± 0.19 k	6.02 ± 0.17 g	6.10 ± 0.16 f	5.90 ± 1.16 j
	100	5.71 ± 0.18 n	5.92 ± 0.16 i	6.10 ± 0.21 f	6.50 ± 0.22 b	6.23 ± 0.18 d
	200	5.70 ± 0.16 n	5.96 ± 0.13 h	6.21 ± 0.19 e	6.75 ± 0.21 a	6.33 ± 0.20 c
Vit C (mg/100 g)	0	218.00 ± 2.54 a	195.8 ± 2.45 e	152.49 ± 3.06 i	125.85 ± 3.35 l	101.88 ± 2.29 p
	50	218.00 ± 2.51 a	200.58 ± 3.01 d	158.57 ± 3.26 h	130.69 ± 2.87 k	106.73 ± 3.74 o
	100	218.00 ± 2.78 a	205.42 ± 2.84 c	163.68 ± 2.98 g	135.77 ± 3.05 j	112.50 ± 3.16 n
	200	218.00 ± 2.64 a	210.98 ± 3.21 b	178.75 ± 3.07 f	149.65 ± 2.64 i	123.34 ± 3.34 m
TPC (mg/kg)	0	181.66 ± 2.14 a	135.73 ± 2.89 e	98.88 ± 2.84 i	60.47 ± 2.85 n	45.33 ± 2.54 p
	50	180.16 ± 2.14 a	145.7 ± 2.54 d	109.65 ± 3.84 h	77.47 ± 2.45 l	54.33 ± 3.01 o
	100	180.56 ± 2.13 a	154.03 ± 2.01 c	118.78 ± 2.89 g	85.47 ± 3.15 k	62.93 ± 2.98 n
	200	181.86 ± 2.15 a	163.53 ± 2.85 b	132.95 ± 4.08 f	95.4 ± 2.54 j	75.23 ± 2.64 m

Mean values with different letter are significantly different by Fisher's least significant (LSD) test ($P < 0.05\%$). Each value is the mean of three replicates.

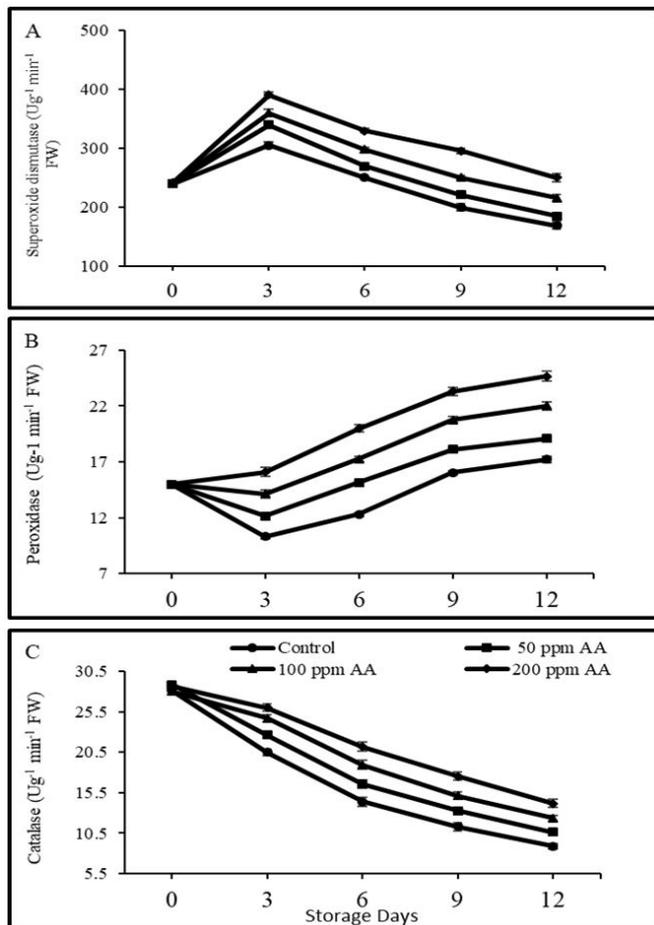


Figure 2. Effect of postharvest application of AA on activities of superoxide dismutase (A), peroxidase (B), and catalase (C) of guava fruit during storage at 25 °C for 12 days. Each value is the mean of three replicates, and vertical bars indicated standard errors.

AA-treated fruits showed 1.27 and 1.09-time greater SOD activity after 12 d of storage, respectively. (Figure 2A). The changes in POD activity decreased first, then linearly increased during storage periods, in all treatments (Figure 2B). Nevertheless, AA-treated fruits showed relatively higher POD activity during storage than control. Likewise, 200 ppm AA and 100 ppm AA treated fruits showed greater POD activity which was 1.43-time and 1.27 time higher, after 12 d of storage than untreated fruits, respectively (Figure 2B). Regardless of treatments, activity of CAT showed gradual decrease during storage phase. However, CAT activity declined noticeably in control than AA-treated fruits. Likewise, 200 ppm AA-treated fruits showed 1.59-fold higher CAT activity than control, after 12 d of storage. (Figure 2C).

Generally, antioxidant defense-enzymes (SOD, POD, and CAT) are considered of great potential for scavenging ROS and, therefore, play a key role in maintaining fruit postharvest quality. In our study, SOD, POD and CAT enzymes tended to have higher activity in AA-treated guava fruits than control, which is consistent with Lin et al. (2007), who reported higher enzymatic activities as well as also found lower concentration of H_2O_2 and less lipid peroxidation. Decrease in decay incidence, reduction in enzymatic activities of SOD, POD and CAT that hinders antioxidant compounds might be associated with capacity of AA to retain fruit quality (Sogvar et al., 2016). However, the molecular mechanism involved in AA enhancing enzyme activities and TPC is unknown and should be explored further.

4 Conclusions

Guava is highly perishable fruit and less work was conducted on the shelf life and postharvest. In this study four concentrations of AA were evaluated on physiochemical and enzymatic changes in guava cv. Golla for 12 d of storage at ambient storage conditions. Our results showed that 200 ppm AA application

significantly reduce weight loss, decay percentage, increased SSC, TA, TS, Vitamin C, and TPC, and decreased pH and SSC: TA ratio of guava fruits during ambient storage. Meanwhile, AA application significantly delayed SOD, POD, and CAT activities, thus reducing oxidative stress. Over all the quality of guava fruits were maintained by application of ascorbic acid during storage as compared to untreated fruits. Our results suggest that ascorbic acid effectively improves the quality attributes and maintained activities of antioxidative enzymes in guava fruits, and would be feasible for guava storage on a commercial use.

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