

Cashmirins A and B, new antifungal and urease inhibitory prenylated coumarins from *Sorbus cashmiriana*

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Cashmirins A (**1**) and B (**2**), new prenylated coumarins, have been isolated from the EtOAc-soluble fraction of the whole plant of *Sorbus cashmiriana* Hedlung, Monog. along with seselin (**3**), scopoletin (**4**), 3-hydroxyxanthyletin (**5**) and luteolin (**6**), reported for the first time from this species. Their structures were elucidated by spectroscopic techniques including MS, 1D and 2D NMR spectroscopy. Both new compounds **1** and **2** were investigated for biological activities and showed significant antifungal and urease inhibitory activities. Compounds **1** and **2** exhibited significant activity against *Aspergillus flavus*, *Macrophomina phaseolina*, *Trichophyton simii*, *Trichophyton schoenleinii*, and *Pseudallescheria boydri*. Both compounds also exhibited significant inhibitory activity against Jack bean urease with IC₅₀ values of 28.2±0.12 µM and 30.3±0.18 µM, respectively compared to thiourea used as positive control.

Keywords: *Sorbus cashmiriana*. Prenylated coumarins. Cashmirins. Antifungal activity. Urease inhibition.

INTRODUCTION

The genus *Sorbus* (Rosaceae) comprises of 200 species which are commonly grown in Asia, Africa and South America. Plants of the genus *Sorbus* are found in customary and local medicines that are used as anti-diarrhea, diuretic, anti-inflammatory, anti-diabetic, vaso-protective, broncho- and vasorelaxant, along with potent antioxidative qualities. In Pakistan, it is represented by seven species. One of these is *Sorbus cashmiriana* Hedlung, Monog, which is a tree of two seasons, one in the spring with lovely pink-tinted flowers and one in the autumn when the leaves are gone, and glorious white fruits shine out. It is distributed in Kashmir and the western Himalayas. A tea made from its bark is

used to treat nausea. The bark preparation is also used to treat heart diseases. Berries are used to cure scurvy (Bhattacharjee, 2003; Perry, Metzger, 1980; Krishna, 1972; Jayaweera, 1982; Krachmal, 1980). Previously, six triterpenes (Kazmi *et al.*, 2007; Kazmi *et al.*, 2009; Kazmi *et al.*, 2011) and two coumarins (Khan *et al.*, 2015) have been reported by us from this species. The ethnopharmacologic and chemotaxonomic importance of the genus *Sorbus* prompted us to carry out further phytochemical studies on *S. cashmiriana*. Herein we report the isolation and structural elucidation of two new prenylated coumarins named as cashmirins A (**1**) and B (**2**) (Figure 1) along with seselin (**3**) (Cazal *et al.*, 2009), scopoletin (**4**) (Darmawan *et al.*, 2012), 3-hydroxyxanthyletin (**5**) (Chen *et al.*, 2010) and luteolin (**6**) (Nissler, Gebhardt, Berger, 2004), reported for the first time from this species.

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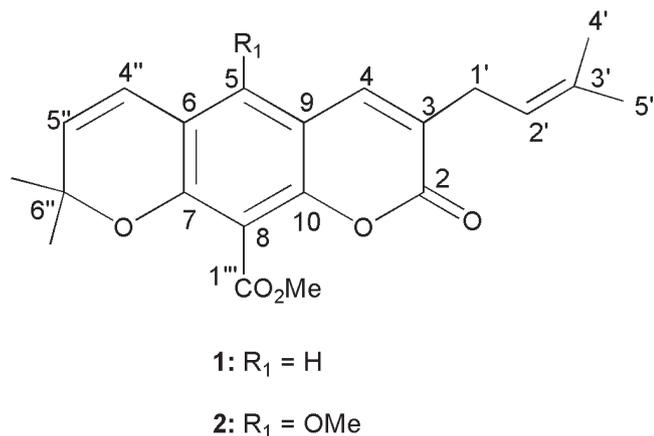


FIGURE 1 - Structures of cashmirins A (1) and B (2).

The plant kingdom has provided a number of therapeutic compounds, such as antifungal, antibacterial, antiparasitic, antihistamine, analgesics, anti-inflammatory, asthma medications and many more (Ficker *et al.*, 2003; Islam *et al.*, 2001; Jones *et al.*, 2000; Omar *et al.*, 2000). Some of the traditionally used plants may lead to the development of new antifungal agents that are in increasing demand due to their resistance to traditional medicines (White *et al.*, 1998).

Urease (EC.3.5.1.5) has been identified as a key virulence determinant in the pathogenesis of a variety of clinical conditions, with implications for human and animal health, as well as agriculture. It leads to the pathogenesis of urolithiasis, pyelonephritis, hepatic coma, and urinary catheter encrustation by being specifically involved in the formation of infection stones (Mobley *et al.*, 1989; Mobley *et al.*, 1995).

In the current study, we have described the antifungal and urease inhibitory activity of the new prenylated coumarins named as cashmirins A (1) and B (2).

MATERIAL AND METHODS

General experimental procedure

The UV and IR spectra were recorded on a Hitachi UV-3200 and JASCO 320-A spectrometers, respectively. NMR data were recorded on a Bruker AV-500MHz spectrometer (500 MHz for ¹H and 125 MHz for ¹³C) in CDCl₃ with tetramethylsilane (TMS) as an internal standard. The chemical shift values are

reported in ppm (δ) and the coupling constant (J) is in Hz. EI- and HREI-MS were recorded on Finnigan MAT 12 and MAT 312 spectrometers in m/z and %. Thin layer chromatography (TLC) was executed on precoated with silica gel F₂₅₄ plates (E. Merck, Darmstadt, Germany): detection at 254 nm and by spraying with ceric sulfate in 10% H₂SO₄ solution. Silica gel (230–400 mesh, E. Merck, Darmstadt, Germany) was used for column chromatography.

Plant material

The whole plant of *Sorbus cashmiriana* Hedlung, Monog was collected from Kashmir and identified by Sher Wali Khan, Department of Biological Science, Karakoram International University Gilgit. A voucher specimen (No. KUH 73/67 760) has been deposited with the Herbarium of the Department of Botany, University of Karachi, Pakistan.

Extraction and isolation

The whole plant *S. cashmiriana* (16 kg) was shade dried ground and extracted with MeOH (3 × 40 L). The MeOH extract was evaporated under reduced pressure to yield a residue (300 g), which was divided into n-hexane (70 g), CHCl₃ (60 g), EtOAc (78 g), and H₂O (80 g) soluble sub-fractions. The EtOAc sub-fraction was subjected to column chromatography over silica gel eluting with n-hexane/CHCl₃ in increasing order of polarity to obtain five fractions F₁-F₅. The fraction F₄ which eluted with n-hexane/CHCl₃ (30:70) was further purified through preparative TLC using n-hexane/CHCl₃ (45:55) to afford compound 6 (18 mg). The fraction F₃ which eluted with n-hexane/CHCl₃ (40:60) was further chromatographed, eluting with n-hexane/CHCl₃ (55:45) to obtain compounds 4 (35 mg) and 5 (30 mg) from the top and the tail fractions, respectively. The fraction F₂ which eluted with n-hexane/CHCl₃ (50:50) was further chromatographed and successively eluted with n-hexane/CHCl₃ (60:40) and (55:45) to obtain compounds 1 (5 mg) and 2 (5.5 mg), respectively. The fraction F₁ which eluted with n-hexane/CHCl₃ (60:40) furnished compound 3 (25 mg).

Cashmirin A (1)

White amorphous solid; m.p. 107-108 °C; UV λ_{\max} (MeOH) 339, 264, 235, 220 nm; IR ν_{\max} (KBr) 1730, 1720, 1635, 1602, 1518 cm^{-1} ; EIMS m/z (rel. int.) 354 (10), 339 (32), 300 (21), 296 (100), 285 (45), 162 (72); HR-EIMS (m/z) 354.1463 $[\text{M}]^+$ (calcd for $\text{C}_{21}\text{H}_{22}\text{O}_5$, 354.1467); ^1H and ^{13}C NMR data, see Table I.

CashmirinB (2)

White amorphous solid; m.p. 107-108 °C; UV λ_{\max} (MeOH) 338, 264, 233, 222 nm; IR ν_{\max} (KBr) 1728, 1722, 1632, 1602, 1522 cm^{-1} ; EIMS m/z (rel. int.) 384 (9), 369 (28), 330 (18), 316 (20), 302 (32), 296 (100), 285 (52), 244 (40), 162 (75); HR-EIMS (m/z) 384.1576 $[\text{M}]^+$ (calcd for $\text{C}_{22}\text{H}_{24}\text{O}_6$, 384.1572); ^1H and ^{13}C NMR data, see Table I.

Antifungal activity assay

The antifungal assay was performed on human, animals and plants pathogens by agar tube dilution method (Ahmad *et al.*, 2007) using ten pathogenic fungi which are illustrated in Table II. The crude extracts, compounds **1**, **2** and the standard drugs (each 200 $\mu\text{g}/\text{ml}$ of Sabour Dextose Agar) were subjected to antifungal activity against *Aspergillus flavus* ATCC 32611, *Aspergillus niger* ATCC 1015, *Macrophomina phaseolina* ATCC 53789, *Trichophyton simii* ATCC 25923, *Microsporium canis* ATCC 36299, *Trichophyton schoenleinii* ATCC 22775, *Fusarium solani* ATCC 36031, *Pseudallescheria boydri* ATCC 44330, *Candida glabrata* ATCC 90030, and *Rhizoctonia solani* ATCC 76131.

Urease inhibitory activity

Reaction mixtures comprising 25 μL of enzyme (Jack bean urease) solution and 55 μL of buffers containing 100 mM urea were incubated with 5 μL of test compounds at 30 °C for 15 min in 96-well plates. Urease activity was determined by measuring ammonia production using the indophenol method (Weatherburn, 1967). 45 μL each of phenol reagent (1% w/v phenol and 0.005% w/v sodium nitroprusside) and 70 μL of alkali reagent (0.5% w/v NaOH

and 0.1% active chloride NaOCl) were added to each well. The increasing absorbance at 630 nm was measured after 50 min, using a microplate reader (Molecular Device, USA). In a final volume of 200 μL , all reactions were performed in triplicate. The results were processed by using Soft Max Pro software (Molecular Device, USA). Assays were performed at pH 8.2 potassium phosphate buffer (0.01 M $\text{K}_2\text{HPO}_4 \cdot 3\text{H}_2\text{O}$, 1 mM EDTA and 0.01 M LiCl). Percentage inhibitions were calculated from the formula $100 - (\text{OD}_{\text{test well}}/\text{OD}_{\text{control}}) \times 100$. Thiourea was used as the standard inhibitor of urease.

RESULTS AND DISCUSSION

Cashmirin A (**1**) was obtained as a white amorphous solid. The molecular formula was established as $\text{C}_{21}\text{H}_{22}\text{O}_5$ through HR-EIMS showing an $[\text{M}]^+$ peak at m/z 354.1463 (calcd 354.1467). The IR spectrum showed the presence of an α , β -unsaturated lactone (1720 cm^{-1}), ester carbonyl (1730 cm^{-1}), aromatic C=C (1602 and 1518 cm^{-1}) and olefinic (1635 cm^{-1}) functionalities. The UV spectrum showed absorption maxima at λ_{\max} 220, 235, 264, and 339 nm, characteristic of 6-substituted-7-oxygenated-pyranocoumarins (Seiji *et al.*, 1989).

The ^{13}C NMR (BB and DEPT) spectra of **1** showed 21 carbon signals, including five methyl, five methine, one methylene and ten quaternary carbons (Table I). The downfield signal at δ_{C} 164.0 and 161.4 ppm were assigned to the carbonyl carbons of the ester and α , β -unsaturated lactone moieties, respectively. The olefinic carbons of the coumarin nucleus were observed at δ_{C} 139.2 and 128.2 ppm. The oxygenated aromatic carbons appeared at δ_{C} 166.7 and 155 ppm. The signals of dimethylchromene moiety were observed at δ_{C} 129.4, 123.0 and 28.0 ppm (2 \times Me), respectively. The signals of tri-substituted olefinic carbons at δ_{C} 133.2 and 121.7 ppm, methyl groups at δ_{C} 25.0 and 18.9 ppm together with methylene carbon at δ_{C} 30.2 ppm could be attributed to γ,γ -dimethylallyl moiety. In the EIMS, the γ,γ -dimethylallyl moiety was further confirmed by a base peak at m/z 285 $[\text{M}-\text{C}_5\text{H}_9]^+$. The signal at δ_{C} 56.4 ppm could be assigned to the methoxyl carbon of the methyl ester.

The ^1H NMR spectrum (Table I) showed the signals of γ,γ -dimethylallyl moiety at δ_{H} 1.75 and 1.62 (3H each,

br. s, Me), 3.23 (d, $J = 7.0$ Hz, 2H) and 5.25 ppm (m, CH). The notable up-field shift of the H-4 signal to δ_{H} 7.29 ppm instead of 7.65–8.03 ppm and the lack of its coupling with H-3, allowed us to assign the γ,γ -dimethylallyl moiety to C-3 (Cordova, Garelli, 1974). It further showed characteristic signals of dimethylchromene ring [δ_{H} 1.38 ppm ($6''\text{-Me} \times 2$) and olefinic protons at δ_{H} 5.49 and 6.41 ppm (d, $J = 9.5$ Hz)]. The spectrum further showed an aromatic proton as singlet at δ_{H} 7.02 ppm and carbomethoxy protons at δ_{H} 3.78 ppm. One of the major fragment at m/z 296 [$\text{M}-59$] $^+$ in EIMS resulted from the loss of methyl carboxylate group. The collective data showed close resemblance to those of previously reported 3-prenylxanthyletin (Cordova, Garelli, 1974), the only notable difference being the presence of methyl carboxylate group at C-8. The location of various substituents was finally confirmed by HMBC correlations (Figure 2). The presence of γ,γ -dimethylallyl moiety at C-3 could be confirmed by 3J correlations of H-1' with C-4 (δ_{C} 139.2 ppm), C-2 (δ_{C} 161.4 ppm) and C-3' (δ_{C} 133.2 ppm) as well as 2J correlations with C-3 (δ_{C} 128.2 ppm) and C-2' (δ_{C} 121.7 ppm). The attachment of 2, 2-dimethylchromene ring at C-6 and C-7 positions of the aromatic ring could be inferred by 2J correlation of H-4'' with C-6 (δ_{C} 116.7 ppm) and 3J correlations with C-7 (δ_{C} 158.7 ppm) and C-5 (δ_{C} 120.7 ppm). The presence of the methyl ester group was evident by the 3J correlation of methoxyl protons (δ_{C} 3.78 ppm) with ester carbonyl (δ_{C} 164.0 ppm). Its presence at C-8 was authenticated by aromatic proton H-5 which showed 3J correlations with C-4 (δ_{C} 139.2 ppm), C-7 (δ_{C} 158.7 ppm), C-10 (δ_{C} 155.0 ppm) and C-4'' (δ_{C} 123.0 ppm), revealing substitution at C-8. All

these data were in complete agreement with the assigned structure of cashmirin A (**1**) as 3-(γ,γ -dimethylallyl)-6'',6''-dimethylpyrano-8-(methylcarboxylate)-8H-pyrano[2'',3'',7,8]-chromen-2-one.

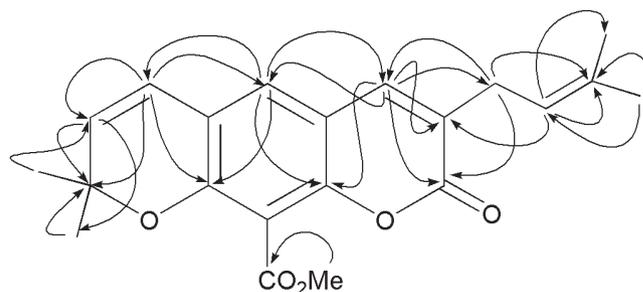


FIGURE 2 - Important HMBC correlations of cashmirin A (**1**).

Cashmirin B (**2**) was obtained as a white amorphous solid. The molecular formula was established as $\text{C}_{22}\text{H}_{24}\text{O}_6$ through HR-EIMS, showing an $[\text{M}]^+$ peak at m/z 384.1576 (calcd 384.1572). The IR and UV spectra were similar to that of **1**. The ^1H and ^{13}C NMR spectra (Table I) were also similar to those of **1** except for an additional signal due to a methoxyl group at δ_{H} 3.89 ppm and δ_{C} 62.2 ppm. The absence of aromatic signal and the downfield shift of C-5 allowed us to assign it to C-5. It was confirmed by HMBC and NOESY spectra. The methoxyl protons at δ_{H} 3.89 ppm showed 3J correlation with C-5 (δ_{C} 151.8 ppm) as well as NOESY interactions of 5-OCH₃ with both H-4 and H-4'', respectively. The structure of cashmirin B (**2**) could thus be assigned as 5-methoxy-3-(γ,γ -dimethylallyl)-6'',6''-dimethylpyrano-8-(methylcarboxylate)-8H-pyrano[2'',3'',7,8]-chromen-2-one.

TABLE I - ^1H and ^{13}C NMR data (500 and 125 MHz) for **1** and **2** (CDCl_3 , δ in ppm, J in Hz)

Position	1		2	
	δ_{H}	δ_{C}	δ_{H}	δ_{C}
2	-	161.4	-	160.8
3	-	128.2	-	128.5
4	7.29 s	139.2	7.64 s	140.0
5	7.09 s	120.7	-	151.8
6	-	116.7	-	110.2
7	-	166.7	-	169.1

TABLE I - ^1H and ^{13}C NMR data (500 and 125 MHz) for **1** and **2** (CDCl_3 , δ in ppm, J in Hz)

Position		1		2	
8	-	111.5	-	100.8	
9	-	109.3	-	101.2	
10	-	155.0	-	156.5	
1'	3.23 d (7.0)	30.2	3.21 d (7.0)	30.4	
2'	5.25 m	121.7	5.26 m	121.5	
3'	-	133.2	-	132.9	
4'	1.62 s	18.9	1.61 s	18.9	
5'	1.75 s	25.0	1.76 s	25.5	
4''	6.41 d (9.5)	123.0	6.53 d (9.5)	118.2	
5''	5.49 d (9.5)	129.4	5.67 d (9.5)	129.7	
6''	-	75.2	-	75.2	
6''-2Me	1.38 s	28.0	1.39 s	28.0	
1'''	-	164.0	-	164.0	
1'''-OMe	3.78 s	56.4	3.79 s	56.2	
5-OMe	-	-	3.89 s	62.2	

The antifungal activity of both **1** and **2** was determined by the agar tube dilution method using ten pathogenic fungi. Both of these showed significant activity against *A. flavus*, *M. phaseolina*, *T. simii*, *T. schoenleinii*, and *P. boydri*; moderate activity against *A.*

niger, *F. solani*, and *C. glabrata* and weak activity against *M. canis*, and *R. solani* (Table II). It is important to note that compound **2** showed slightly more potency than **1** which is probably is due to the presence of additional methoxyl group at C-5.

TABLE II - *In vitro* fungicidal bioassay of crude extract and cashmirins A (**1**) and B (**2**)

Organisms	Inhibition (%) of crude extract	Inhibition (%)		Standard drugs	Inhibition (%) of standard drugs
		1	2		
<i>A. flavus</i>	68	78	80	Amphotericin-B	100
<i>A. niger</i>	50	54	62	Miconazole	90
<i>M. phaseolina</i>	58	69	72	Miconazole	100
<i>T. simii</i>	67	80	82	Ketoconazole	100
<i>M. canis</i>	35	16	25	Benlate	90
<i>T. schoenleinii</i>	62	75	78	Ketoconazole	100
<i>F. solani</i>	48	57	60	Benlate	100
<i>P. boydri</i>	60	74	78	Nystain	90
<i>C. glabrata</i>	58	60	65	Miconazole	100
<i>R. solani</i>	24	11	17	Ketoconazole	100

The inhibitory activity of compounds **1** and **2** against Jack bean urease was determined by the method described in the experimental section. The IC_{50} values of compounds **1** and **2** were found to be $28.2 \pm 0.12 \mu\text{M}$ and $30.3 \pm 0.18 \mu\text{M}$, respectively, as against IC_{50} value of $22.4 \pm 0.15 \mu\text{M}$, which was observed for thiourea that was used as a positive control (Table III).

TABLE III - *In vitro* inhibition of urease of cashmirins A (1) and B (2)

Compound	$IC_{50} \pm \text{S.E.M} (\mu\text{M})$
1	28.2 ± 0.12
2	30.3 ± 0.18
Thiourea	22.4 ± 0.15

S.E.M.: standard error of the mean (n = 3-5).

CONCLUSION

In conclusion, two new prenylated coumarins, named as cashmirins A (**1**) and B (**2**) have been isolated from *Sorbus cashmiriana* Hedlung, Monog. along with four known compounds. Their structures were elucidated by spectroscopic techniques including MS, 1D and 2D NMR spectroscopy. Both new compounds showed significant antifungal activity against *A. flavus*, *M. phaseolina*, *T. simii*, *T. schoenleinii*, and *P. boydri*. The results showed compound **2** was slightly more antifungal than **1**. Both of these compounds also exhibited significant inhibitory activity against the enzyme urease.

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Received for publication on 07th June 2021
Accepted for publication on 20th November 2021