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# PHENOL DEGRADATION IN AN ANAEROBIC FLUIDIZED BED REACTOR PACKED WITH LOW DENSITY SUPPORT MATERIALS

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**Abstract** - The objective of this research was to study phenol degradation in anaerobic fluidized bed reactors (AFBR) packed with polymeric particulate supports (polystyrene - PS, polyethylene terephthalate – PET, and polyvinyl chloride - PVC). The reactors were operated with a hydraulic retention time (HRT) of 24 h. The influent phenol concentration in the AFBR varied from 100 to 400 mg L<sup>-1</sup>, resulting in phenol removal efficiencies of ~100%. The formation of extracellular polymeric substances yielded better results with the PVC particles; however, deformations in these particles proved detrimental to reactor operation. PS was found to be the best support for biomass attachment in an AFBR for phenol removal. The AFBR loaded with PS was operated to analyze the performance and stability for phenol removal at feed concentrations ranging from 50 to 500 mg L<sup>-1</sup>. The phenol removal efficiency ranged from 90-100%. *Keywords*: Phenol; Anaerobic fluidized bed reactor; Biofilm; Polymeric particles.

### INTRODUCTION

Phenol and phenolic compounds are present in wastewater from industries such as coal gasification, production, manufacture chemicals, pharmaceuticals, pesticides, fertilizers, and dyes, and pulp and paper processes (Tay et al., 2001, Veeresh et al., 2005). Phenol and phenolic compounds are also used in the preparation of synthetic resins, antiseptics, dyes, biocides, photographic chemicals, and more (Sá and Boaventura, 2001, Fang et al., 2004). Wastewater that contains phenols and other toxic compounds requires careful treatment before discharge into a receiving water body. Phenol is either toxic (reduces enzymatic activity) or lethal to fish at relatively low concentrations. It imparts objectionable tastes to municipal drinking water, especially chlorinated water, even at very low concentrations (Hosseini and Borghei, 2005). Brazilian legislation has established a phenol concentration of 0.5 mg L<sup>-1</sup> as the limit for wastewater discharge into natural water bodies or municipal sewerage systems. For drinking water, the World Health Organization prescribed a guideline concentration of 0.001 mg L<sup>-1</sup> (WHO, 1994).

Recovery of phenols is economical at high concentrations by physical processes (e.g., solvent extraction or activated carbon adsorption). At intermediate concentrations, i.e., phenol concentrations ranging from 5 to 500 mg L<sup>-1</sup>, biological oxidation

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techniques are feasible (Veeresh *et al.*, 2005). Phenol is biodegradable both aerobically (Sá and Boaventura, 2001, Hosseini and Borghesi, 2005, Tziotzios *et al.*, 2005) and anaerobically (Tay *et al.*, 2001, Fang *et al.*, 2004, Wang *et al.*, 1986, Bolaños *et al.*, 2001, Farooqi *et al.*, 2007).

The anaerobic degradation of phenol offers the advantages of having no oxygen requirement, producing low waste biomass, and generating a valuable waste product, methane gas. Different reactors have been evaluated for the anaerobic treatment of wastewater, both in suspended cultures (e.g., anaerobic digesters) and immobilized ones (e.g., anaerobic filter, expanded bed, and fluidized bed) (Rajeshwari *et al.*, 2000). There has been an increasing interest in the immobilization of microorganisms on inert supports, since this strategy often leads to a more efficient process.

The removal of phenols has been accomplished using anaerobic filters (Khan *et al.*, 1981), expanded bed reactors (Wang *et al.*, 1986), and horizontal flow anaerobic immobilized biomass (HAIB) reactors (Bolaños *et al.*, 2001). However, anaerobic treatment of phenol-containing wastewater has generally been carried out using UASB reactors (Tay *et al.*, 2001, Veeresh *et al.*, 2005, Fang *et al.*, 2004, Farooqi *et al.*, 2007).

The advantage of using anaerobic fluidized bed reactors (AFBR) for the treatment of hazardous waste with inhibitory or recalcitrant compositions has been recently demonstrated (Alvarez et al., 2006, Perez et al., 2007, Haroun and Idris, 2009). One factor that contributes to the efficiency of the fluidized bed process is the minimal liquid film diffusional resistance, which is attributed to the particle motion and liquid velocities. Another factor is the initial dilution of the influent, which provides alkalinity and some neutralization, reduces substrate concentration (important for high COD (chemical oxygen demand) wastes), and contributes to reduction of the shock effect of toxicant spikes (Hickey and Owens, 1981, Iza, 1991).

One of the most important aspects in the design of an AFBR is the choice of support material. The selection of appropriate support material should consider several aspects aside from those related to fluidization; experimental trials to determine the best material are unavoidable. Other considerations relate to the cost of the material and its physical properties (size, shape, particle density, hardness, rugosity and surface area) (Marin *et al.*, 1999).

The particles most often used as supports in fluidized bed reactors are sand and activated charcoal. Sand is a cheap material and insensitive to abrasion.

Meanwhile, activated charcoal is more expensive and requires care with respect to abrasion because turbulence in the system can easily reduce the charcoal in size. Since fluidization of conventional supports usually requires high pump energy, the use of a low density support, such as polymeric particles, previously subjected to chemical treatment, can also improve upon the inherent advantages of fluidized bed bioreactors (Tavares *et al.*, 1994, Tavares *et al.*, 1995, Saucedo-Terán *et al.*, 2004).

However, it is critical that the particulate media used be conducive to a rapid and extensive colonization of firmly attached biomass to assure that stable AFBR performance is attainable shortly after startup. Although the mechanisms and substances involved in adhesion and biofilm formation are not completely known, most studies have emphasized that extracellular polymeric substances (EPS) are the main material responsible for the structural and functional integrity of biofilms due to the cohesive forces they exert, which are responsible for keeping cells together in the form of biofilms, flocs and sludge (Tavares *et al.*, 1995, Flemming and Wingender, 2001, Qureshi *et al.*, 2005).

To date, few systematic studies have compared the startup and steady state performances of AFBR containing different types of support media under similar operating conditions. Therefore, the present study focused on phenol removal by a mixed culture growing on polystyrene, polyethylene terephthalate, or polyvinyl chloride as support materials in AFBR. The biofilms formed on the support materials were evaluated based upon a quantification of the biomass and extracellular polymers. The effects of phenol concentration on the performance of the AFBR and its stability were also investigated.

#### MATERIALS AND METHODS

#### Synthetic Wastewater and Inoculum

The reactor was operated with a synthetic substrate containing phenol as the sole carbon source. The phenol concentrations for treatment ranged from 50 to 500 mg L<sup>-1</sup>, plus the following nutrients (in mg L<sup>-1</sup>): NH<sub>2</sub>CONH<sub>2</sub>, 62.5; NiSO<sub>4</sub>·7H<sub>2</sub>O, 0.5; FeSO<sub>4</sub>·7H<sub>2</sub>O, 2.5; FeCl<sub>3</sub>·6H<sub>2</sub>O, 0.25; CaCl<sub>2</sub>·2H<sub>2</sub>O, 23.5; CoCl<sub>2</sub>·6H<sub>2</sub>O, 0.04; SeO<sub>2</sub>, 0.035; KH<sub>2</sub>PO<sub>4</sub>, 42.5; K<sub>2</sub>HPO<sub>4</sub>, 10.85; Na<sub>2</sub>HPO<sub>4</sub>·7H<sub>2</sub>O, 16.7; NaHCO<sub>3</sub>, 1,000; and yeast extract, 300. The inoculum was obtained from the anaerobic sludge of a UASB reactor treating effluent from a swine slaughterhouse (volatile suspended solids (VSS) ~ 20 g L<sup>-1</sup>).

## **Evaluation of Anaerobic Phenol Degradation in Batch Reactors**

The experiments for evaluating the potential of the mixed culture for anaerobic phenol degradation were performed in a shaking incubator. Duran bottles of 2 L, with a working volume of 1 L, were used as anaerobic batch reactors. Besides phenol and nutrient additions, 10% (v/v) of inoculum (total volatile solids (TVS) ~ 960 mg L<sup>-1</sup>) was fed into the Duran bottles; the initial pH was 6.8. The headspace of the bottle was purged with N<sub>2</sub> (100%) for 60 seconds. The rubber topped bottles were set in a shaking incubator adjusted to 30°C and 150 rpm. The concentrations of phenol studied were 137 and 355 mg L<sup>-1</sup>. The experiments were performed in duplicate.

## **Support Material for Immobilization of the Anaerobic Sludge**

Particles of polystyrene (PS), polyethylene terephthalate (PET), and polyvinyl chloride (PVC) were used in the AFBR as support materials for biomass immobilization. The PS, PET and PVC particles were submitted to a prior chemical treatment to improve their surface characteristics (rugosity, porosity and electrical charge) as reported by Tavares et al. (1994, 1995) and Saucedo-Terán et al. (2004). This procedure provided better conditions for the microorganism's attachment. This chemical treatment consisted of submerging PS and PET particles in a sulfochromic acid solution for 50 minutes, rinsing them in water, submerging them in concentrated nitric acid for 20 minutes, rinsing them in water again, and finally drying them in an oven at 40°C (Tavares et al., 1994, 1995). However, this procedure was applied four times for the polystyrene particles, identical to the procedure used by Sancinetti (2004), who observed the gradual increase in the presence of particles roughness under an optical and scanning microscope.

PET particles suffered the same treatment as polystyrene particles however there was a 30% mass loss of these particles, without the need of another acid bath. After dipping in sulphochromic acid, the PVC particles became soft and dark colored. It was therefore decided to submerge them in concentrated nitric acid for 20 minutes only once. The PVC particles were rinsed in water and dried in an oven at 40°C. The main characteristics of the support materials are shown in Table 1.

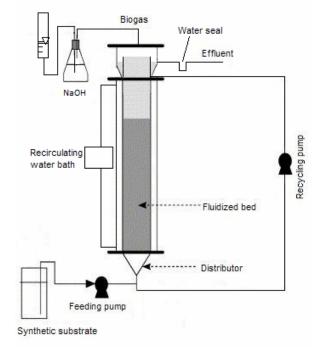
**Table 1: Characteristics of the support materials** 

	PS	PET	PVC
Size (mm)	2.5 x 2.5	3.0 x 3.0	3.0 x 3.0
Density (g cm <sup>-3</sup> )	1.05	1.25	1.11
Shape	cylinders	cylinders	cylinders
Weight of support	190	226	200
material loaded (g)			
Height of fixed bed (cm)	34	30.5	31
Minimum fluidization	0.74	1.35	1.16
velocity (cm s <sup>-1</sup> )			

#### **Anaerobic Fluidized Bed Reactors**

Figure 1 shows the schematic of the three identical reactors used for evaluating the low density supports in AFBR. The reactors used for the experimental work were constructed of an 80.0 cm long, 3.5 cm internal diameter acrylic tube. This inner tube was enclosed in an outer jacket through which water was circulated to maintain the temperature of the reactor at  $30 \pm 1^{\circ}\text{C}$ . Three sampling points were installed along the reactor length to obtain liquid and bioparticle samples. At the top of the reactor, a stainless steel screen was installed to prevent the escape of particles. A liquid displacement system was used to measure the amount of biogas generated, as shown in Figure 1.

The useful volume of the reactor with polystyrene particles was 606 cm<sup>3</sup>, with PET particles was 332 cm<sup>3</sup> and in the reactor with PVC particles was 578 cm<sup>3</sup>.



**Figure 1:** Schematic representation of the AFBR.

#### Startup and Operational Conditions of the AFBR

The three reactors were packed with different support materials (PS, PET, or PVC), inoculated, and started up simultaneously under identical conditions. The AFBR were fed with synthetic wastewater containing phenol (100 mg L<sup>-1</sup>), nutrients, and inoculum adapted to phenol for approximately 40 days at a concentration of 460 mg TVS L<sup>-1</sup>. The headspace of the AFBR was flushed with N<sub>2</sub> (100%) for 180 seconds. The reactors remained at rest for 18 hours. After that, the AFBR were maintained under total recycle for seven days to promote the growth and attachment of biomass on the support materials; they were then switched to continuous mode with a designated HRT of 24 h. Effluent recycle was maintained at a rate corresponding to 1.3 times the minimum fluidization flow rate of each support material in order to achieve completely mixed conditions within the reactor. The total volatile solids concentration after seven days of recirculation was 580 mg TVS L<sup>-1</sup>, indicating an increase of the microbiological populations during the immobilization phase. The influent phenol concentration varied from 100 to 400 mg L<sup>-1</sup>

After selecting the best polymeric particle, one system was operated to analyze the performance and stability of the AFBR for phenol removal. To simplify the analysis of the results, the research was divided into eight experimental phases, corresponding to different values of phenol concentration for an HRT of 24 h. When a steady state was reached, the phenol concentration was increased progressively from 50 to approximately 500 mg L<sup>-1</sup>. The reactor was operated for 116 days in total.

#### **Chemical and Microbiological Analyses**

Phenol analysis was performed using the 4aminoantipyrine colorimetric method (APHA, 1998). COD, TVS, VSS and pH values were measured according to the Standard Methods (APHA, 1998). Alkalinity, as CaCO<sub>3</sub>, was determined as described by Dillalo and Albertson (1961) and adapted by Ripley et al. (1986). Volatile fatty acids (VFA) concentrations were assessed using chromatography (HP 6890/FID) equipped with a 30 m long HP INNOWAX column, with an internal diameter of 0.25 mm and a film thickness of 0.25 µm. The injector temperature was kept at 250°C; the oven was maintained at 100°C for 3 min, after which it was heated at a rate of 5°C/min to 180°C and kept at that temperature for 5 min (Moraes et al., 2000).

Structural analysis of biofilm samples was performed by using a Zeiss DSM-960 digital scanning microscope (Varesche *et al.*, 1997). Biomass adhesion to the polymeric particles was determined according to the methods of Chen and Chen (2000). Quantification of the extracellular polymeric substances (EPSs) in protein form was performed in accordance with the method proposed by Lowry *et al.* (1951) and modified by Peterson (1977), using bovine serum albumin as a standard. Analysis of EPSs in the carbohydrate form was carried out according to the methods of Dubois *et al.* (1956) using lactose as a standard.

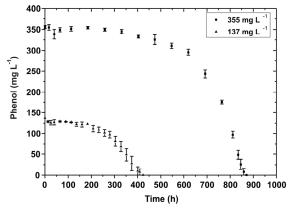
#### RESULTS AND DISCUSSION

#### **Evaluation of Anaerobic Phenol Degradation by** the Mixed Culture in Batch Reactors

As mentioned above, the condition of 10% sludge in batch reactors was adopted for the initial experiments, and the initial biomass concentration was 960 mg TVS L<sup>-1</sup>. As can be seen from Figure 2, the results accompanied total phenol consumption over 423 hours (17 days) and 871 hours (36 days) for the tests with 137 and 355 mg phenol L<sup>-1</sup>, respectively. Figure 2 shows the average value of two experiments for each phenol concentration and the deviation of the data. The solids concentrations in the batch reactors increased to 140 and 220 mg TVS L<sup>-1</sup> for phenol concentrations of 137 and 355 mg L<sup>-1</sup> respectively, which indicated the formation of an active biomass in the batch reactors. The final pH was within the normal range of anaerobic system operation, between 7.2-7.4 (Speece, 1996).

Increased acetic acid production was observed in experiments with concentrations of 137 and 355 mg phenol L<sup>-1</sup>. As can be seen in Figure 3, the acetic acid remained at the end of the bioassay and was beginning to be consumed; it is likely that, given time, all of the acetate would have been converted into methane. The degradation of phenol by fermentative methanogenic consortia is well established. Phenol is first carboxylated to p-hydroxybenzoic acid. This compound is next dehydroxylated to benzoic acid, which is then reduced, cleaved and transformed into acetate and propionate and ultimately methane (Ramakrishnam and Gupta, 2006).

Therefore, the gradual adaptation of the microbial consortium to phenol for a period of approximately 40 days in the anaerobic batch reactors secured the establishment of inoculum for the startup of the AFBR.

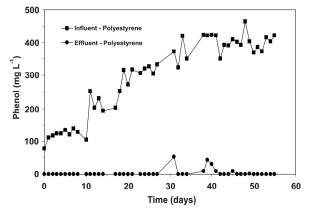


**Figure 2:** Temporal variation of the phenol concentration.

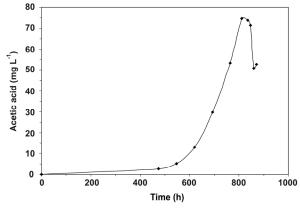
## Effect of Phenol Concentration on the Performance of AFBR Packed with Polymeric Particles

After the adaptation of the inoculum to phenol in anaerobic batch reactors, 1.3 L of adapted sludge supplemented with nutrient solution was used to start up each of the AFBR. As can be seen in Figures 4 to 6, the startup of the AFBR was fast. Steady state conditions were reached after three days of continuous operation at 100 mg L<sup>-1</sup>, with a phenol removal efficiency of 100%. The average removal of phenol was close to 100% at an influent phenol concentration of 400 mg L<sup>-1</sup> with each of the three support materials tested.

For the AFBR containing PVC, an instability appeared in the phenol removal efficiency on the 45<sup>th</sup> day, which remained until the end of the continuous operation of this reactor. This instability was attributed



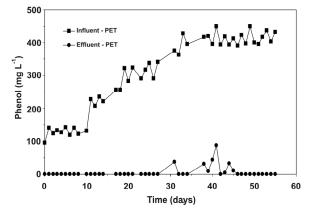
**Figure 4:** Variation of the phenol concentration in the AFBR containing PS over the operating period.



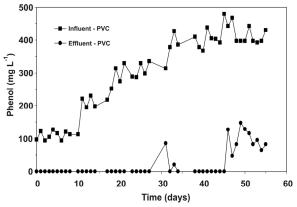
**Figure 3:** Temporal variation of acetic acid production for the experiment with 355 mg phenol L<sup>-1</sup>.

to the deformation of the PVC particles from the 10<sup>th</sup> day onward, which caused increases in the roughness and size of the support material, favoring microbial adhesion. The deformation of the PVC particles led to bed stratification in the AFBR and increased biofilm thickness; bed stratification has many negative effects on fluidized bed reactor performance.

Thicker biofilms pose diffusion limitations and washout problems. Stratification can be attributed to the influence of a biofilm on the settling velocity of the particles. The presence of a biofilm coating decreases the overall particle density, thereby increasing its buoyancy. The biofilm also increases the particle's size, thereby increasing the drag force exerted on it by the liquid flowing upward. The particles in a fluidized bed are expected to segregate according to size and mean density (Saravanan and Sreekrishnan, 2006).



**Figure 5:** Variation of the phenol concentration in the AFBR containing PET over the operating period.



**Figure 6:** Variation of the phenol concentration in the AFBR containing PVC over the operating period.

The results obtained for the AFBR containing PVC were in accordance with those of Hidalgo and García-Encina (2002), who observed that a decrease of the expansion provoked a biomass increase in the reactor and a decrease in the removal rate of organic matter. This can be explained by a mass transfer limitation phenomenon, because substrate utilization rates, related to diffusional resistance, are strongly dependent on the reactor mixing intensity. Volatile fatty acids accumulated within the system when the expansion of the fluidized bed was lower than 10%.

The EPS development on the support materials did not affect the stability of the AFBR containing PS and PET, leading to slight increases in the bed heights of these AFBR. Figures 7 to 9 show the variations in TVS contents, EPS contents in the form of carbohydrates, and EPS contents in the form of proteins, respectively, in the biomass attached to PS, PET, and PVC particles as a function of continuous

operation time of the AFBR. Attached biomass was higher for PVC particles, while EPS content in carbohydrates and proteins was greater for polystyrene particles. The initial decrease in the TVS/support ratio in the AFBR can be attributed to the batch operation mode during the first seven days to favor biomass growth and attachment on the support materials before switching to the continuous mode; after this initial period, there was an increase in the attached biomass in the AFBR.

The higher EPS content in the PVC particles compared to the PS and PET particles could be related to the thicker biofilm and bed stratification. In the AFBR containing PS and PET, neither particle deformation nor bed stratification occurred. However, these reactors had slight increases in the attached biomass from the 20<sup>th</sup> day of continuous operation onward, while the EPS contents in the AFBR containing PS and PET remained almost constant.

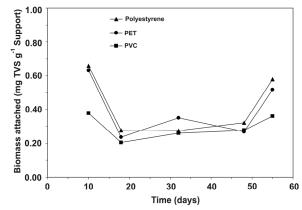
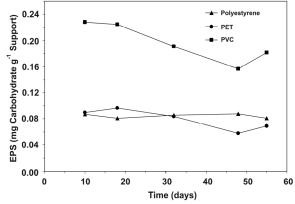


Figure 7: Attached biomass in the AFBR.



**Figure 8:** EPS content in the carbohydrate form in the AFBR.

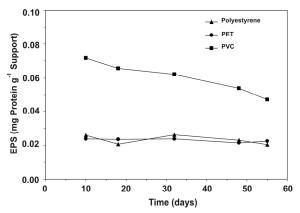


Figure 9: EPS content in the protein form in the AFBR.

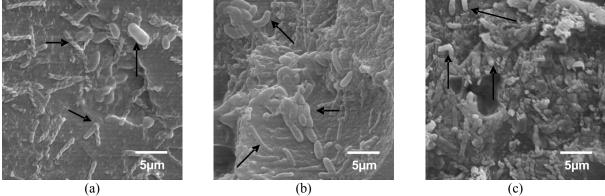
According to Kuroda *et al.* (1988), the formation of the biofilm depends not only on the characteristics of the biomass and fluid regimes, but also on the characteristics of the solid supports. However, in biofilm reactors, detachment is the most important process for balancing growth and removal of the excess biomass. Detachment controls the solid residence time of the microorganisms, which in turn influences the overall performance and stability of the bioreactor (Derlon *et al.*, 2008).

Microbiological characterizations of the biofilms on the support materials were performed periodically through scanning electron microscopy (SEM) micrographs. Figure 10 shows the predominant presence of straight, oval, and spiral bacilli, vibrio, and cocci in the AFBR containing polymeric particles. Figure 10a shows the predominance of rods adhered to the PS particles. On the PET particles (Figure 10b), the presence of straight rods, oval rods, and curved rods also were observed. Figure 10c also shows bacteria with morphologies similar to rods and cocci adhering to the PVC particles.

## Stability of Phenol Removal in AFBR Packed with Polystyrene Particles

Although the AFBR packed with PS and PET particles showed similar good results for phenol degradation (organic loading removals of 0.84 and 0.98 kg phenol m<sup>-3</sup> d<sup>-1</sup>, respectively), the authors of this study chose PS as the best support material for biomass attachment in AFBR for phenol removal because it has the lowest density (Table 1), reducing the energy costs for pumping.

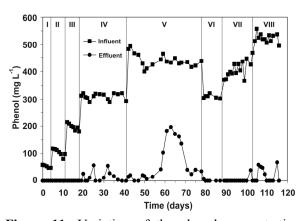
The 100% removal efficiency confirmed that the reactor startup was fast and that a stationary state had been reached after three days of continuous operation at an influent phenol concentration of 50 mg L<sup>-1</sup>. The fast reactor response at the beginning of the operation was attributed to the fact that the system was kept as a closed circuit for the first seven days to promote microbial adhesion to the PS particles. The results obtained in this study were similar to the startup periods previously reported for biofilm reactors.



**Figure 10:** SEM micrographs of biomass attached to support material on the 55<sup>th</sup> day of continuous operation: (a) PS, (b) PET, and (c) PVC (magnification 5,000X).

In an expanded bed reactor containing granular activated charcoal (Wang *et al.*, 1986), the startup period was eight days for a phenol concentration of 358 mg L<sup>-1</sup>. In a HAIB reactor packed with polyurethane foam (Bolaños *et al.*, 2001), the startup period was 33 days for a phenol concentration of 50 mg L<sup>-1</sup>, with an average removal efficiency of 69%. In a UASB reactor (Faaroqi *et al.*, 2007), the startup period was 40 days for a phenol concentration of 200 mg L<sup>-1</sup>, with an average removal efficiency of 83%.

Figure 11 shows the results obtained for influent and effluent phenol concentrations for all operational phases of the AFBR containing PS. Notably, the phenol removal efficiency was 100% for average influent phenol concentrations of 51.6 to 197.1 mg L<sup>-1</sup> (operation phases I to III). When the concentration was increased to 308.5 mg L<sup>-1</sup> (phase IV), the efficiency was 97% until the 25<sup>th</sup> day. On the 25<sup>th</sup> and 32<sup>nd</sup> days, the phenol removal efficiencies decreased to 82% due to operational problems of the pumping systems, which increased the influent flow rate and thereby caused shock loads in the reactors. However, the AFBR quickly recovered their maximal phenol removal efficiency after organic load reduction through adjusting the influent flow rate to a value corresponding to an HRT of 24 h. When the average influent phenol concentration in the AFBR was increased to 445.4 mg L<sup>-1</sup> the phenol removal efficiency ranged from 91-100%.



**Figure 11:** Variation of the phenol concentration with time in the PS reactor.

In order to test the stability of the AFBR containing PS, the influent flow rate was increased by 100% (decreasing the HRT to 12 h) on the 56<sup>th</sup> day and maintained for 48 h with an average influent phenol concentration of 440.3 mg L<sup>-1</sup> (phase V). As can be seen in Figure 12, the phenol removal efficiency of the reactor decreased to 58%; on the 71<sup>st</sup> day, it returned to 91%. Over the period from the

71st to the 78th day, an increasing amount of suspended biomass was observed in the effluent of the AFBR. According to Perez et al. (2007), in biofilm systems the amount of suspended biomass is negligible with respect to the attached biomass. This aspect allows low HRT levels to be maintained with respect to the suspended systems, and this causes the washing out of the suspended biomass (the fluidized bed retains the growth support media in suspension by drag forces exerted by upflowing wastewater). Therefore, the levels of suspended solids in the effluent arising from the detachment of biomass would be expected to be higher during periods when the applied organic loading rate was increased, whether due to an increase in influent flow rate or to a decrease of HRT. To ensure the complete recovery of the AFBR between days 79 and 87 (phase VI), the influent phenol concentration was reduced to an average of 307.8 mg L<sup>-1</sup>. Phenol removal efficiency returned to 100%. As can be seen in Figure 11, during phase VII (89th to 103rd day), the average influent phenol concentration was increased to 408.1 mg L<sup>-1</sup>, and the phenol removal efficiency remained at 100%. From the 103<sup>rd</sup> day on, the average influent phenol concentration was increased to 523.4 mg L<sup>-1</sup>, and phenol removal efficiency ranged from 89-100%.

The results obtained in this study indicated good performance for the AFBR containing PS particles up to phenol concentrations of approximately 500 mg L<sup>-1</sup>. The use of inoculum from swine manure previously adapted to phenol contributed positively to the fast AFBR startup, high phenol removal efficiencies, and reactor stability towards hydraulic loading shocks. The values of the phenol removal efficiencies were satisfactory and consistent with removal efficiencies of phenol in expanded bed reactors (Wang *et al.*, 1986), HAIB reactors (Bolaños *et al.*, 2001), and UASB reactors (Farooqi *et al.*, 2007) with influent phenol concentrations up to 500 mg L<sup>-1</sup>.

The COD reduction efficiencies were 70% for phase I, 87% for phase II, 93% for phase III, and 91% for phase IV. After the end of the hydraulic loading shock, the COD reduction efficiency returned to 95-100%.

High COD reduction efficiency indicates that phenol was actually removed from the liquid phase not just converted to intermediate products. The VFA were detected by chromatography during reactor operation with 440.3 mg phenol L<sup>-1</sup> (phase V). On day 58, 15.48 mg L<sup>-1</sup> of acetic acid was measured and on day 61 69.28 mg L<sup>-1</sup> of acetic acid, 21.11 mg L<sup>-1</sup> of propionic acid and 4.22 mg L<sup>-1</sup> of

isovaleric acid were measured. The volatile acids presence on day 61 confirms the results of phenol concentration that indicated a decrease of reactor efficiency to 58%. The analysis for the next period, phase VI, after the decrease of phenol concentration to 307.8 mg L<sup>-1</sup>, did not indicate volatile acids, proving the reactor recovery and system stability. During the operation of an anaerobic expanded bed reactor with activated carbon as the support material. Wang et al. (1986) observed that acetic acid was the only acid measured in the effluent and that its concentration rose with the increase of feed phenol concentration. For phenol concentrations up to 703 mg L<sup>-1</sup>, the effluent acetic acid concentration was lower than 2 mg L<sup>-1</sup>; nonetheless, for phenol concentrations of 1,492 mg L<sup>-1</sup> and 2,959 mg L<sup>-1</sup>, the acetic acid concentrations were 22 mg L<sup>-1</sup> and 374 mg L<sup>-1</sup>, respectively.

Even with the decrease in reactor efficiency during phases IV and V, effluent pH variation was not observed. The pH stability may be due to the alkalinity generation, the absence of acidity generation, and the phenol removal efficiency, as phenol is a weak acid. The average values for effluent pH during the reactor operation were 8.61 for phase I, 8.5 for phase II, 8.6 for phase III, 8.1 for phase IV, 8.1 for phase V, 8.2 for phase VI, 8.0 for phase VII, and 7.9 for phase VIII. Fatty acid accumulation was not observed, probably due to phenol and COD usage and the values of pH and alkalinity observed.

Figure 12 shows the results for bicarbonate alkalinity concentrations at all phases of reactor operation. An initial decrease of effluent alkalinity concentration is evident from phase I until day 7 of phase II. This decrease can probably be attributed to maintenance of a closed system during the first seven days of the startup process. Even though the reactor nominally reached steady state after three days of

continuous operation, the alkalinity stabilization occurred only in phase III. After phase II, median alkalinity concentration was 201.8 mg L<sup>-1</sup>; a gradual increase of effluent alkalinity concentration occurred in the other phases. The values obtained were 422.8 mg L<sup>-1</sup> for phase III, 522.7 mg L<sup>-1</sup> for phase IV, 521.8 mg L<sup>-1</sup> for phase V, 550.0 mg L<sup>-1</sup> for phase VI, 577.0 mg L<sup>-1</sup> for phase VIII.

Only in phase I was the standard deviation higher than the average value of the effluent alkalinity, 590.0 mg L<sup>-1</sup>. This anomaly was due to the effluent alkalinity instability at the beginning of continuous reactor operation. Alkalinity was presumably generated because the effluent concentrations presented higher values than the influent alkalinity in all phases of reactor operation. Even after the hydraulic loading shock, the reactor operation at all phenol concentrations was stable with respect to bicarbonate alkalinity.

SEM micrographs of polystyrene particles at the end of each operational phase showed straight rods, oval rods and cocci (Figure 13). In this work, the microorganisms of the Domain *Bacteria* were responsible for the organic matter (phenol, organic acids such as propionate, butyrate and formate) degradation and, consequently, for the formation of acetic acid, that probably was used by acetoclastic Archaea (*Methanosaeta*).

The adopted methodology of acid pre-treatment of the polystyrene particles before being put inside the reactor was effective, as were the startup and reactor inoculation processes. An increase in microorganism adhesion to the surface and cavities of the polystyrene particles with operating time was observed in SEM micrographs. SEM micrographs confirmed that polystyrene particles can be used as support material for biofilm formation and development in AFBR.

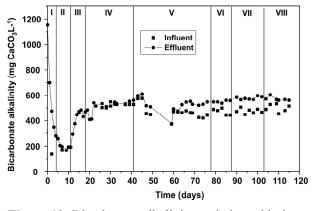
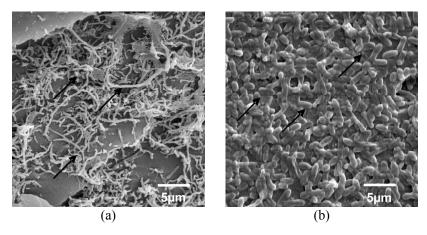


Figure 12: Bicarbonate alkalinity variation with time.



**Figure 13:** SEM micrographs: cell adhesion and biofilm formation on polystyrene particles. (a) Bacilli observed during phase V, instability of the reactor (magnification 3,000X). (b) Bacilli forming polymeric matrix during phase VII (the end of experiment; magnification 5,000X).

#### **CONCLUSIONS**

The AFBR containing PS, PET, and PVC operated with removal efficiencies close to 100% at phenol concentrations up to 400 mg L<sup>-1</sup>. The methodology used to start the reactor was effective and the inoculum from swine manure proved to be a good source of microorganisms to perform anaerobic phenol degradation. PVC particles showed more attached extracellular polymeric substances than PET or PS; these substances caused stratification and loss of stability of the reactor after 40 days of operation. This phenomenon was attributed to deformation of the PVC particles during AFBR operation. Reactors filled with PET and PS particles showed stable performance; both have appropriate characteristics as supports in an AFBR. The reactor filled with PET particles showed better results in the degradation of phenol in relation to the reactor filled with PS. However, PS presents the best overall characteristics for the operation of AFBR reactors, especially its lower density, which reduces the energy cost of the pumping needed for large scale fluidization.

The reactor operated for a long period of time with phenol as the sole carbon source. Previously adapted inoculum contributed positively to the rapid response observed in the reactor, with high removal efficiencies for phenol and COD. In this test, the PS particles proved to be suitable for microbial adhesion and maintenance of the biological process, as indicated by the presence of straight rods, oval rods and cocci, which were associated with phenol removal. The anaerobic fluidized bed reactor yielded

good performance with respect to phenol and COD removal from the synthetic wastewater used. The efficiency of phenol consumption was higher than 95% and COD reduction was higher than 85% for phenol concentrations up 500 mg  $\rm L^{-1}$ .

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#### **NOMENCLATURE**

anaerobic fluidized bed	
reactors	_
chemical oxygen demand	mg L <sup>-1</sup>
extracellular polymeric	
substances	
flame ionization detector	
horizontal flow anaerobic	
immobilized biomass	
reactor	
hydraulic retention time	h
polystyrene	
polyethylene terephthalate	
polyvinyl chloride	
scanning electron	
microscopy	
total volatile solids	mg L <sup>-1</sup>
upflow anaerobic sludge	
	reactors chemical oxygen demand extracellular polymeric substances flame ionization detector horizontal flow anaerobic immobilized biomass reactor hydraulic retention time polystyrene polyethylene terephthalate polyvinyl chloride scanning electron microscopy total volatile solids

blanket reactor

mg L<sup>-1</sup>

mg L<sup>-1</sup>

VFA Volatile fatty acids
VSS volatile suspended solids

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