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Comparison of Single-Step Methods to Enrich Lipase Concentrations in Bacterial Cell Lysates

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HIGHLIGHTS

- Lipases were enriched from *Geobacillus* and *Anoxybacillus* lysates
- For each lipase, ethanol precipitation resulted in the highest enrichment fold
- Ethanol precipitation is a simple enrichment method that can result in high yields

Abstract: Lipases are currently used in food technology for the modification of fats and oils. The thermal stability of lipase is an essential characteristic for this application. This study compares four individual single-step methods (heat treatment, ethanol precipitation, ammonium sulfate precipitation, and size-exclusion chromatography) to enrich lipase concentrations from thermophilic bacterial (*Geobacillus stearothermophilus* and *Anoxybacillus flavithermus*) cell lysates. SDS-PAGE and size exclusion chromatography were used to determine the molecular weights of the lipases and the enrichment efficiencies were determined using specific enzyme activities. The molecular weight of *G. stearothermophilus* lipase was approximately 42 kDa, and approximately 33 kDa for *A. flavithermus* lipase. For each organism, ethanol precipitation resulted in the highest enrichment fold, followed by ammonium sulfate precipitation, gel filtration and heat treatment respectively. The highest yields for *G. stearothermophilus* lipase were obtained with ammonium sulfate precipitation, followed by gel filtration, and ethanol precipitation respectively. The highest yields for *A. flavithermus* lipase were obtained from heat treatment followed by ammonium sulfate precipitation, gel filtration and ethanol precipitation respectively. Ethanol precipitation and heat treatment are simple methods for enzyme enrichment from cell lysates and can result in high enzyme yields with moderate enrichment folds compared to complex multi-step purification methods.

Keywords: thermophilic lipase, enzyme enrichment, zymography

INTRODUCTION

Enzymes are considered nature's catalysts. Currently, the fermentation of microorganisms is commonly used to produce enzymes [1]. Esterases (EC. 3.1.1.3, carboxylic ester hydrolases) and lipases (EC. 3.1.1.1, triacylglycerol hydrolases) are two major classes of hydrolytic enzymes of importance [2]. They hydrolyze the fatty acids in triacylglycerols in the presence of water. Additionally, they act as catalysts in interesterification

and synthesis reactions [3]. Lipases can synthesize, hydrolyze, and rearrange fatty acids, thus, resulting in their extensive use in industrial applications [3].

Lipases play an important role because of their biocatalysts and biotechnological applications. Commercial lipases are used in the pharmaceutical, biodiesel and food science fields [4]. Furthermore, lipases are used in food technology for the modification of fats and oils to change triacylglycerol fatty acid composition by interesterification [5, 6]. Most of these industrial enzymes are of microbial origin and are used in the modification or esterification of fats and oils to improve the overall health value, sensory and functional attributes of the oils [7].

Lipases are produced from several sources including plants, animals, and microorganisms (bacteria, yeast, and fungi) and can be isolated and purified for use in food processing. The high-catalytic activity, yield, and ease of purification makes the use of lipases very advantageous. Microbial lipases have been isolated from different sources including *Thermomyces lanuginosa*, *Pseudomonas sp.*, *Rhizopus oryzae*, *Asperigillus sp.*, *Candida sp.* and *Bacillus sp.* Lipases are characterized by the ability to complete reactions of both hydrolysis and synthesis, as well as working with different substrates such as alcohols, amines, and carboxylic acids [8]. In general, the thermal stability of the enzyme is an important characteristic that is desired by the food industry for interesterification of oils [9]. Because of this characteristic, the use of heat tolerant lipases can increase the reaction rate, stabilize the substrate, reduce microbial contamination, and improve the solubility of fat [10]. Thermophilic *Geobacillus* and *Anoxybacillus* are bacteria that can withstand high temperatures, ranging from 50-90 °C and produce lipases that may have important industrial applications [9].

Lipase may be purified from crude extracts using a variety of techniques depending on the characteristics of the enzyme, such as molecular mass and isoelectric point. The purpose of this process is to increase the purity of the lipase and remove detrimental components including proteases [7]. Purification of enzymes has traditionally been achieved by several techniques such as precipitation, hydrophobic interaction chromatography, gel filtration, and ion exchange chromatography. Often a combination of these techniques are used. Here, we compared individual, single-step methods for the enrichment of lipases from *G. stearothermophilus* and *A. flavithermus* cell lysates.

Recently, lipases were purified from thermophilic bacteria but with several purification steps to get a 3% yield. The sequential steps including ammonium sulfate precipitation, Phenyl Sepharose CL4B, followed by Sephadex G-100 chromatography [11]. Also, a lipase was purified using a combination of heat treatment, DEAE-Cellulose, followed by Sephadex G-150 and G-25 to obtain a 19.7% purification yield [12].

To date, there are few studies which have investigated the use of ethanol precipitation and heat treatment for the enrichment of lipases from thermophilic microbes. In addition, lipases have not been enriched using a single-step to get a 50% yield. In this study, we compared the efficiency of four different single-step methods, precipitation by ethanol, ammonium sulfate, heat treatment, or size-exclusion chromatography to enrich lipases from two different thermophilic bacteria (*G. stearothermophilus* and *A. flavithermus*). The enrichment yield and fold were determined for each method.

MATERIAL AND METHODS

Sephadex G-100, 4-nitrophenyl acetate, sodium dodecyl sulfate (SDS), tetramethyl ethylenediamine (TMED), acrylamide/bis-acrylamide 29:1, ammonium persulfate, glycine, Coomassie Brilliant Blue R-250, and bromophenol blue were purchased from Sigma-Aldrich (USA). Triton X-100, ammonium sulfate, and glycerol were obtained from Mallinckrodt Specialty Chemicals Co. (USA). Tris HCL and tris base were obtained from Fisher Scientific (USA). Protein concentration was determined using the BCA kit from Thermo Scientific (USA).

Growth of microorganisms

The culture of *G. stearothermophilus* was prepared by adding 0.1 ml of stock solution obtained from NAMSA *G. stearothermophilus* spore suspension (2.4×10^6 spores/0.1 ml, biological indicator for STEAM, LOT: S90601, Fisher Scientific, USA) to 10 ml of sterile water. This diluted stock was incubated in water bath for 10 min at 80°C. One ml of this was moved to 25 ml of tryptic soy broth (TSB) media in a sterile 250 ml Erlenmeyer flask covered with sterile foil and incubated at 55°C for 24 h in a shaker at 100 rpm. The optical density (OD) of bacteria growth was measured using a spectrophotometer (BioSpec-1601, Shimadzu, USA) at 600 nm [13].

A. flavithermus TNO-09.006 culture was a gift from Remco Kort (Top Institute Food and Nutrition, The Netherlands) and grown using 0.1 ml culture mixed with 25 ml TSB with 0.05 g potato starch in a sterile 250

ml Erlenmeyer flask covered with sterile foil and incubated at 55° C for 14-18 h in a shaker at 100 rpm. Growth of *A. flavithermus* was confirmed by measuring OD 600 spectrophotometrically [14].

Frozen stocks were prepared by inoculating 2 ml of an overnight culture to 20 ml of a 30% glycerol solution (glycerol/TSB, w/v) and stored in 2.0 ml cyro-vials at -20°C. Frozen stocks (0.1 ml) of each organisms were added to 25 ml TSB in a sterile 250 ml Erlenmeyer flasks covered with sterile foil and incubated at 55°C for 16-18 h in a shaker at 100 rpm [13]. The 25 ml growth of bacteria was added to 1 L of TSB and cells were grown at 55°C at 100 rpm until the OD at 600 nm was between 0.7 to 0.9. The bacteria were separated from the medium by centrifuging at 3000 rpm for 10 min at 10°C. The cells were washed in 10 ml Tris-HCL buffer and centrifuged again as described above. After centrifugation, the cells were mixed with 5-10 ml Tris buffer (50 mM Tris HCL, pH 7.5), and this mixture was sonicated at 40% amplitude in an ice bath (Qsonica, LLC, Newtown, CT) for 1 min using a 4.5 mm microtip and a Qsonica Sonicator power source. The sonicated solution was centrifuged at 10,000 rpm at 0°C for 20 min and the lysate was filtered using a microfilter 1.0 µm (Glassfiber Prefilter, Tullagreen, Carrigtwohill Co. Ireland). The crude protein lysates were used for further experiments.

Total protein and enzyme activity

Lipases were assayed using the substrate 4-nitrophenyl acetate (Sigma-Aldrich, USA). Five ml of substrate stock solutions were prepared using an acetonitrile and isopropanol mixture (1:4 v/v) as previously described [15]. One ml of the stock was diluted with 20 ml of substrate buffer (50 mM Tris HCL, pH 7.5 containing 1mM CaCl₂ and 0.3% Triton X100). Samples were added to substrate buffer and the absorbance at 410 nm was recorded every 5 seconds over 1 min using spectrophotometer at 22°C. One unit of lipase activity was defined as 1 µmol substrate hydrolyzed per minute under the assay conditions. This is referred to as the standard lipase assay. Specific activity was determined by calculating the ratio of the lipase activity (U) and protein concentration (mg) and expressed as U/mg. The protein content was determined using protein determination kit (Pierce Chemical Co., USA).

Gel electrophoresis

SDS-PAGE was carried out using a 5% stacking gel and a 12% running gel under reducing conditions as previously described [16]. Samples containing 20 µg of protein in 20 µl buffer were mixed with 10 µl of SDS sample buffer (3X=1M Tris HCL pH 6.8, 10% SDS, 0.3% Bromophenol blue, 1% β-mercaptoethanol, and 30% glycerol) and heated at 90°C for 5 min. Running buffer, 0.1% (w/v) SDS dissolved in 25 mM Tris and 192 mM glycine pH 8.3, was used at 60 V for the first 15 min and 100 V for an additional 90 min using a Mini-Protean Tetra Cell (Bio-Rad Laboratories, USA). Gels were stained using 0.1% (w/v) Coomassie Brilliant Blue R-250, 45% (v/v) methanol and 10% (v/v) acetic acid for 20 min. A low molecular range protein standard was used as molecular weight markers (β-galactosidase 120 kDa, bovine serum albumin 90 kDa, ovalbumin 50 kDa, carbonic anhydrase 34 kDa, β-lactoglobulin 26 kDa, and lysozyme 20 kDa) (Thermo Scientific Prestained Protein Molecular Weight Marker # 26612) [16]. The migration distances of the standards was graphed vs the log molecular weight of the standards. The migration distances of the putative lipase bands was determined and the molecular weights were calculated.

In this study, zymography was used in conjunction with SDS-PAGE under reducing conditions with both chromogenic zymography and transfer zymography. For chromogenic zymography, after electrophoresis, the SDS gel was washed for 10 min in 2.5% (v/v) Triton X-100 (50 mM Tris HCL pH 8) and other 10 min with 1% (v/v) Triton X-100 in the same Tris buffer. The gel was washed with 20% isopropanol for 20 min [17]. Then washed with distilled water three times and the gel was incubated with an activation buffer (50 mM Tris-HCL pH 8.0) for 30 min at room temperature. The washing steps were used to remove denaturing agents to refold the lipases. The gel was then placed in the chromogenic substrate solution (0.01%w/v) phenol red, 10 mM CaCl₂ with lipidic substrate (0.5% (v/v) glyceryl tributyrat (Sigma-Aldrich, USA) and 0.5% (v/v) olive oil, then incubated at 45 to 55 °C. A yellow band observed in the gel was indicative of lipase activity [18].

After electrophoresis, the denaturing agents were removed from gel using same solutions as described above in the chromogenic zymography for transfer zymography. The substrate gel was prepared as described in Table 1. A sandwich blot was assembled and included filter paper, SDS gel, substrate gel, filter paper and was loaded into Bio-Rad Mini Trans-Blot electrophoretic transfer cell unit. Running buffer without SDS was used (25 mM Tris and 192 mM glycine pH 8.3) and the transfer voltage was 15 V for 25 min at 4 °C [19]. The gel was removed and incubated with activation buffer (50 mM Tris HCL, 25 mM sodium chloride, pH 8.9) at 55 °C for 24 h. Lipase activity was observed as white bands under a grayish background [19, 20].

Table 1. Substrate gel for transfer zymography

Component	Volume (mL)	Final concentration (%)
40% Acrylamide/Bis-acrylamide (29:1 %) ratio	2.4	40
Glycerol Tributryrate	0.72	12
Olive oil	0.147	2.5
Distilled Water	2.67	44.5
Sonication one minutes (5 pulse at 20% amplitude)		
Ammonium persulfate (10% w/v)	0.06	1
TEMED	0.003	0.05

Enrichment of lipases from cell lysates

Lipases from *G. stearothermophilus* and *A. flavithermus* extracts were enriched in cell lysates using four single-step methods including heat treatment, ethanol precipitation, ammonium sulfate precipitation, and size-exclusion chromatography. For each method, protein content and enzyme activity were measured using the standard lipase assay and protein determination kit and protein profiles were determined by SDS-PAGE analysis. Initially, cells from a 25 mL overnight growth were used for each method, then when the optimal conditions were determined, cells from 2 L of an overnight growth were used to determine the enrichment fold and yield. All experiments were replicated 3 times and all assays were conducted in triplicate and standard deviations (SD) were calculated.

The cell lysates from *G. stearothermophilus* and *A. flavithermus* after heating at (50-80 °C) for 10 min in a water bath was done and the temperature was monitored every 25 sec to ensure the required temperature was maintained using a precision digital thermometer. Samples were placed in an ice bath after heat treatments to reduce the temperatures. The solutions were centrifuged at 10,000 x g for 30 min at 4°C to precipitate denatured proteins [21] and the supernatants were saved. The supernatants were assayed for lipase activity and protein content using the standard lipase assay and protein determination kit described above.

The cell lysates were mixed with various concentrations of ethanol (20-80%). One ml of lysate was diluted with 0 to 7 ml with deionized water before cooling to 0°C in an ice bath. Ethanol (2 to 9 ml) was gradually added to the lysates to get final ethanol concentrations ranging from 20-80% at 4°C [16]. Samples were centrifuged at 8,000 x g at 4°C for 20 min and the precipitate (pellet) was washed in 50 mM Tris HCL buffer, pH 7.5, centrifuged again with the same conditions, then solubilized in 1.0 mL of the Tris buffer. The solubilized pellets were assayed for lipase activity and protein content using the standard lipase assay and protein determination kit described above.

Ammonium sulfate protein precipitation was conducted as described by a previous method with some modifications [11]. Different concentrations of ammonium sulfate (0, 20%, 40%, 60%, 80%, and 95%) were added to cell lysates at 4°C with stirring. Each pellet, after centrifugation at 10,000 x g for 30 min at 4°C, was collected and dissolved in 50 mM pH 7.5 Tris HCL buffer at 4°C. The solutions were dialyzed using dialysis cassettes (Slide-A- Lyzer 10,000 MWCO, Thermo Scientific, USA) for 24 h at 4°C with fresh dialysis buffer replacing the buffer every 4 h. After dialysis, samples were assayed for lipase activity and protein content using the standard lipase assay and protein determination kit described above.

The cell lysates were loaded onto a Sephadex G-100 column (36.0 cm x 2 cm) equilibrated with 50 mM Tris buffer, pH 7.5. Proteins were then eluted with the same buffer with a flow rate of 0.5 mL/min. Protein fractions were collected using an ISCO Model 1200 (Teledyne ISCO, USA) fraction collector. The fractions were assayed for lipase activity and protein content using the standard lipase assay and protein determination kit described above. Additionally, protein standards (bovine serum albumin or BSA at 67 kDa), pepsin (34.5 kDa), trypsin (23.3 kDa), lysozyme (14.3 kDa) and blue dextran were run through the column and their elution volumes recorded. The elution volumes of the standards vs the molecular weights were plotted and the elution volumes of the lipase activity peaks were used to calculate the estimated molecular weights.

RESULTS

Identification of lipases in cell lysates

Thermophilic bacteria grow at high temperature range of 45-70 °C [22] and produce thermophilic enzymes. Lipases from two different thermophilic bacteria, *G. stearothermophilus* and *A. flavithermus*, were enriched in cell lysates using 4 different single-step methods. The total lipase activity and total protein in the cell lysates from *G. stearothermophilus* were 4761 U and 117.6 mg with a specific activity of 40.5 U/mg using

4-nitrophenyl acetate as the substrate (Table 2). Cell lysates from *A. flavithermus* showed a total activity of 803 U, and the total protein was 110.1 mg with a specific activity of 7.3 U/mg using the same substrate (Table 2). Overall, the results show that lysates from *G. stearothermophilus* have a specific activity that is 5.5 times higher than the activity of lysates from *A. flavithermus*. There was a slight difference in the total protein content of lysates from each organism.

Table 2. Enrichment efficiency of lipases from thermophilic bacteria within each single-step method

Source	Method	Total Activity (U)	Total protein (mg)	Specific activity (U/mg)	Enrichment (fold)	Yield (%)
<i>Geobacillus stearothermophilus</i>	crude	4761 ± 58	117.6 ± 1	40.5 ± 0.6	1	100
	Heat (70°C)	2572 ± 4	56.0 ± 0.4	46.0 ± 0.4	1.1	54
	Ethanol (80%)	2411 ± 120	37.8 ± 0.4	63.8 ± 3	1.5	50
	Amm (95%)	3918 ± 148	73.9 ± 0.9	53.0 ± 2	1.3	82
	Gel filtration	3430 ± 251	62.6 ± 2	54.8 ± 7	1.3	72
<i>Anoxybacillus flavithermus</i>	crude	803 ± 28	110.1 ± 6	7.3 ± 0.1	1	100
	Heat (55°C)	730 ± 23	95.3 ± 2	7.6 ± 0.1	1	91
	Ethanol (60%)	430 ± 3.0	10.0 ± 0.1	43.0 ± 0.2	5.9	53
	Amm (80%)	680 ± 13.2	65.0 ± 0.9	10.4 ± 2.0	1.4	84
	Gel filtration	538 ± 34.8	42.6 ± 1.2	12.6 ± 1.4	1.7	67

In comparison to previous findings, the total lipase activity of lysates from *G. stearothermophilus* (4761 U) was higher than that of *Geobacillus* sp strain ARM and [23], *Geobacillus stearothermophilus* AH22, [12]. Also, the specific activity was higher than that of *Geobacillus stearothermophilus* PS 11 [24]. The total activity and specific activity of lysates from *Anoxybacillus flavithermus* was 803 U and 7.3 U/mg; which is lower than the lipase activity reported from *Anoxybacillus flavithermus* HBB 134 [11].

SDS-PAGE was used to analyze the cell lysates from the thermophilic bacteria after denaturation under reducing conditions to estimate the molecular weights of the lipases. Figure 1A shows the protein profile of lysates from *G. stearothermophilus*; the arrow indicates the putative lipase band with the molecular weight of approximately 42 kDa. Figure 2A shows the protein profile of lysates from *A. flavithermus*; the arrow indicates the putative lipase band with the molecular weight of approximately 33 kDa.

Activity staining is a method used to verify the bands and molecular weights of enzymes. Activity staining was performed using a substrate (glyceryl tributyrates and olive oil) gel electrophoresis with lysates from thermophilic bacteria. There are two different techniques used to indicate lipase activity, chromogenic and transfer zymography. For both methods, the proteins were reactivated by washing the SDS-PAGE gel with activation buffer to remove SDS and the gel was incubated at 55°C for 1-2 h in TrisHCL buffer pH 8.0 to renature the lipases. This research used both zymography techniques to confirm the molecular weights.

In chromogenic zymography, an active lipase will reduce the pH of the gel, which changes the phenol red color to yellow [18]. Figures 1B and 2B show a yellow band since the lipase changed the color in the gel from red to yellow corresponding to a lipase band. The lipase band in lysates from *G. stearothermophilus* have a molecular weight of approximately 42 kDa (Figure 1B) and that from *A. flavithermus* has an approximate molecular weight of 33 kDa (Figure 2B) indicated by the arrows.

Transfer zymography was also used to determine the molecular weights of the lipase bands. The lipase band was transferred from the SDS-PAGE gel to the substrate gel using electroblotting. The substrate gel was then incubated with activation buffer and a color change indicates the presence of an active lipase [17]. Figures 1C and 2C show zones of lipase activity on the white or cloudy background of lysates from *G. stearothermophilus* with molecular weight of approximately 42 kDa (Figure 1C) and *A. flavithermus* with molecular weight of approximately 33 kDa (Figure 2C). As shown in Figure 1C, the lipase band was clearly visible for *G. stearothermophilus* because a higher lipase activity was observed after incubating the gel with reaction buffer. Whereas, the contrast between the band and the background was less visible for lipase isolates from *A. flavithermus* because the lipase activity was lower, perhaps because the lipase did not renature as well as the lipase from *G. stearothermophilus* (Figure 2C).

The molecular weight of enzymes obtained were different for each microbe which agrees with previous reports that show that the molecular weight of lipases are different and are dependent on the source. For example, the molecular weight of a lipase from *Bacillus gladioli* ATCC10248 was 43 kDa [25], *Bacillus*

thermoleovorans CCR11 was 11 kDa [26], *Bacillus subtilis* NRRL B8079 was 54 kDa [27], and *Bacillus acidocaldarius* ORF3 was 34 kDa [27]. Specifically, for the *Geobacillus* and *Anoxybacillus* genus, the molecular weight of lipase from *Geobacillus* sp. SBS-4S was 43.4 kDa [28] *Geobacillus stearothermophilus* JC was 44.3 kDa [29], *Anoxybacillus* sp. PDF1 was 26 kDa [30], and *Anoxybacillus flavithermus* DSM was 24.36 kDa [31].

Enrichment of lipases from cell lysates

In this study, we compared the efficiency of four different single-step methods including heat treatment, ethanol precipitation, ammonium sulfate precipitation, and size-exclusion chromatography (Sephadex G-100) to enrich lipases from the thermophilic bacterial lysates. Enrichment fold and yield from the four methods were used to compare the results of each method. A high yield with a high enrichment fold indicates a greater concentration of the protein of interest with removal of contaminants including other proteins present in the crude cell lysates.

Enrichment by heat treatment results in precipitating proteins via centrifugation which aggregate during the heat treatment. The thermotolerance of some proteins will keep them in solution while less heat tolerant proteins will aggregate and precipitate with centrifugation. In this study, different temperatures were used to treat crude cell lysates. The activity of lysates from *G. stearothermophilus* remained high until 75°C, while the lowest enzyme activity was at 80 °C (Figure 3A). Comparing the soluble protein with the lipase activity, we selected 70°C to use for the heat treatment since it showed both high soluble protein and high lipase activity. The lipase activity from *A. flavithermus* was active when treated at 50 to 80°C; the enzyme activity was highest at 55°C and also showed high soluble protein at that temperature (Figure 3B), therefore, 55°C was used for the heat treatment method.

Figure 4, lanes 4A2 and 4B2 show the supernatants of lysates from each organism treated at 70 and 55°C. Heat treatment of cell lysates from *G. stearothermophilus* (seen in Figure 4A, lane 2) can be compared with the untreated crude lysate (lane 1). The number of protein bands at 70°C were less than seen in the lysates. This indicated that heat treatment at 70°C precipitated proteins which are not thermostable and increases the enrichment fold. The protein bands from the untreated lysate and the heat treated lysate at 70°C include a band with the molecular weight of 42 kDa. Additionally, the lysate and heat treated lysate using 55°C for *A. flavithermus* showed a band with a molecular weight of approximately 33 kDa (Figure 4B lane 2).

The enzymatic yield and enrichment fold after heat treatment at 70°C for *G. stearothermophilus* was 54% and 1.1-fold respectively while that from *A. flavithermus* was 91% and 1 fold at 55°C (Table 2). The specific activity of lysates from *G. stearothermophilus* after treat at 70°C was 46 U/mg which was 6 time higher than the specific activity from *A. flavithermus* after treatment at 55°C (7.6 U/mg). The total protein in *G. stearothermophilus* lysates was reduced by almost half, while there was little reduction in the total protein of *A. flavithermus* lysates. For *G. stearothermophilus* this result refers to removal of 46% of proteins from the lysates using one single enrichment step and an increase in lipase specific activity to 1.1 (Table 2). Ekinci *et al.* [12] reported an 85% yield and 1.3-fold purification of a lipase from *G. stearothermophilus* AH22 using 70°C. Also, heat treatment was used to purify a lipase from *Staphylococcus aureus*, and the yield was 74.6%, and the purification fold was 1.57-fold using 55°C [32].

The enrichment of lipases from thermophilic bacteria lysates was evaluate using ethanol precipitation. Ethanol precipitation is a technique used to reduce the protein content through precipitate of protein and miscibility with water. This precipitation technique is an easier method compared to ammonium sulfate precipitation because the precipitate does not need an additional dialysis step to remove residual salts [16]. Temperature control is essential with this method to avoid protein denaturation so the temperature must be held at below 10°C.

To determine the best ratio of ethanol concentration for precipitating lipases, different concentrations of ethanol (20-80% (v/v)) were used. The lipase activity from *G. stearothermophilus* (pellet) increased with an increase in ethanol up to 80% ethanol. The protein content of the pellets was also the lowest at 80% ethanol (Figure 5A). With 20-70% ethanol, the lipase partially precipitated, and the enzyme activity was approximately 35-40 % less than at 80% ethanol. Protein patterns of the 80% ethanol precipitate is shown in the SDS-PAGE (Figure 4A lane 3) and a band at 42 kDa is observed.

Ethanol precipitation of lysate from *A. flavithermus* is shown in Figure 5B. The highest lipase activity was observed at 60% ethanol with a low protein concentration. With increased ethanol concentration to 80%, the lipase activity in the pellet decreased by approximately 50%, which indicates that the concentration of ethanol over 60% precipitated the lipase. The protein patterns of lysates and ethanol precipitated protein at 60% (v/v)

is shown in the SDS-PAGE and the protein band with molecular weight of 33 kDa is observed (Figure 4B lane 3).

Regarding the lipase enrichment by ethanol precipitation, the yield of lipase in the pellet from *G. stearothermophilus* was 50%, and enrichment fold was 1.5-fold (Table 2). Lipase from ethanol precipitation of *A. flavithermus* lysate showed a 53% yield with a 5.9 enrichment fold (Table 2). As a result, ethanol precipitation enriched the lipases with the highest specific activity and enrichment fold compared to other methods for both organisms. In this study, activity yield for thermophilic bacteria was lower than that from *Candida cylindracea* which was 84% and *Penicillium camembertii* Thom PG-3 which was 85.8% using 80% ethanol [33].

Protein precipitate by ammonium sulfate is a classical technique used for the purification of proteins; it is depended on the protein concentration as well as the temperature, pH, and isoelectric point of proteins. Ammonium sulfate, with its low cost and availability, makes it an accessible method for protein enrichment. This study used gradual graded ratios of ammonium sulfate (20-95%) to precipitate protein in the lysates of both organisms while keeping the temperature of the solution below 10.0°C.

For both organisms, an increase in ammonium sulfate resulted in higher specific activities (Figure 6A and B). Ammonium sulfate precipitation (Figure 6A) from *G. stearothermophilus* lysates increased in specific activity and was the highest at 95% ammonium sulfate. At this concentration of ammonium sulfate, the lowest protein concentration was also observed in both 6A and 6B. For lysates from *A. flavithermus* the highest specific activity was observed at 80% ammonium sulfate which also had a low protein concentration. Protein patterns of ammonium sulfate precipitation at 95% (pellet) with lysates from *G. stearothermophilus* are shown in SDS-PAGE (Figure 4A, lane 4). A major protein band with a molecular weight 42 kDa was predominant in the gel. The protein patterns *A. flavithermus* at 80% ammonium sulfate are shown in the SDS-PAGE (Figure 4B lane 4) and with a band at 33 kDa is visible.

From these results, the yield of lipase from *G. stearothermophilus* lysates at 95% ammonium sulfate was 82% with a 1.3 enrichment fold, for *A. flavithermus* there was an 84% yield with a 1.4 enrichment fold (Table 2). Ammonium sulfate precipitation did increase the enrichment fold of lipases from both organisms. The enrichment yield of the lipase from *G. stearothermophilus* was similar to that from *G. stearothermophilus* AH22 lysates which had an 85% yield with a 1.3 fold purification [12]. In this study, the enrichment fold of lipase from *A. flavithermus* was approximately higher (1.4-fold) than lipase from *A. flavithermus* HBB 134 (1.2-fold), and the yield was similar when using an ammonium sulfate concentration between 20 and 60% [11].

Purification of lipases has involved the use of classic chromatographic techniques including ion exchange, gel filtration or affinity chromatography, which usually result in high purification folds. Lipases from thermophilic bacteria were enriched by size-exclusion chromatography (Figure 7A and 7B). All the fractions were assayed for protein concentration and lipase activity. The lipase activity of fractions 20-40 for *G. stearothermophilus* was high with a corresponding high protein concentration. The lipase activity of fractions 60-80 from *A. flavithermus* showed high lipase activity with a relatively low protein concentration. Fractions from the Sephadex G-100 column with lipase activity were pooled and were analyzed by SDS-PAGE (Figure 4A and 4B lanes 5). As expected, the *G. stearothermophilus* peak contained the targeted protein with a molecular weight of 42 kDa, while *A. flavithermus* peak had enriched lipase band with molecular weight of 33 kDa. Additionally, molecular weight protein standards were used to calibrate the column (BSA, pepsin, trypsin, and lysozyme) and their elution times are superimposed on the chromatograms in Figure 7. Based on the elution times of the standards, the molecular weights of the lipases from *G. stearothermophilus* and *A. flavithermus* were calculated to be 42 and 33 kDa, which corroborates with the molecular weight determinations from SDS-PAGE.

The specific activity of lipase from *G. stearothermophilus* in pooled fractions was 54.8 U/mg, which is higher than the lysate (40.5 U/mg) and the total protein decreased approximately 53% from the total protein in the lysate (Table 2). The yield was 72% and the enrichment fold was 1.3. The specific activity of the lipase in pooled fractions from *A. flavithermus* was 12.6 U/mg with a 67% yield and 1.7 enrichment fold (Table 2). This method resulted in a high a enrichment fold for *A. flavithermus*. In previous research, the yield of lipase purified from *Microbacterium* sp. was 20.8% and 2.1 fold using Sephadex G-100 [46]. Also, two different Sephadex resin were used to purify a lipase from *G. stearothermophilus* AH 22; the purification yield was 34.7% with Sephadex G-150 while the yield for a second loading was 19.7% using Sephadex G-25 [12]. While purification yield and purification fold of lipase isolated from *A. flavithermus* HBB 134 were 3% and 7.4-fold using a Sephadex G-100 column [11].

The enrichment efficiency of lipases from *G. stearothermophilus* and *A. flavithermus* with the different single-step methods is summarized in Table 2. Ethanol showed the highest enrichment folds for lipases from both organisms with similar enrichment yields. The enrichment folds using ammonium sulfate were similar for each organism and the enrichment fold using chromatography was higher for the lipase from *A. flavithermus* than from *G. stearothermophilus*. In comparison between both thermophilic sources, ethanol precipitation was the most efficient method with high enrichment folds and yields. Ammonium sulfate and gel filtration techniques showed high lipase yield from both thermophilic bacteria. With each enrichment method, we did not achieve 100% lipase purification and we believe that the putative lipase bands shown in Figure 4 may not be solely lipase proteins. The purpose of this research was to find a simple, single-step method that resulted in an increase in the enrichment fold while maintaining a high enzyme yields.

CONCLUSIONS

Depending on the application of the thermophilic lipase, a method should be selected based on purification need vs. method complexity. The commercial use of lipases in the food industry does not always require high purity [12] while other uses, such as pharmaceutical, do require high purity. Our study enables researchers to choose the desired level of lipase purity, thereby reducing the complexity of the purification process. In our study, lipases were enriched using single-step methods (heat treatment, ethanol precipitation, ammonium sulfate precipitation, and gel filtration chromatography) in cell lysates. Lipases were enriched from two different thermophilic bacteria (*G. stearothermophilus* and *A. flavithermus*) and each method was evaluated for enrichment efficiency by measuring specific enzyme activities. Among the methods, the lipases from *G. stearothermophilus* and *A. flavithermus* enriched by ethanol precipitation had yields of 50% to 53% and enrichments of 1.5-fold and 5.6-fold respectively. Lipases are currently used in food technology; lipase can hydrolyze and esterify fatty acids, thus, resulting in their extensive use in industrial applications.

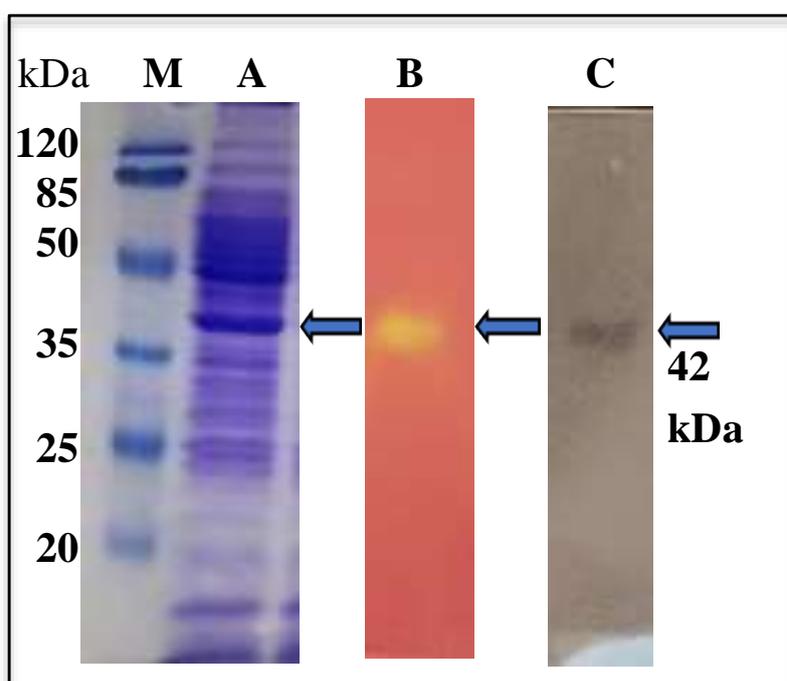


Figure 1. SDS-PAGE analysis of crude cell lysates from *Geobacillus stearothermophilus*. **A.** SDS-PAGE gel stained with Coomassie blue along with molecular weight markers (**M**). **B.** Chromogenic zymography activated with phenol red, glycerol tributyrates and olive oil as substrates. **C.** Transfer zymography activated with glycerol tributyrates and olive oil as substrates.

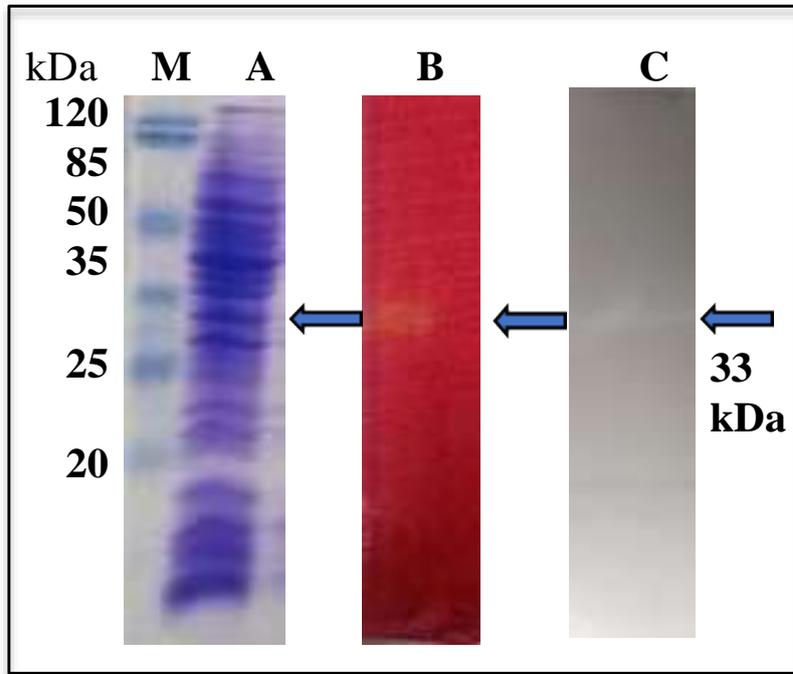


Figure 2. SDS-PAGE analysis of crude cell lysates from *Anoxybacillus flavithermus*. **A.** SDS-PAGE gel stained with Coomassie blue along with molecular weight markers (**M**). **B.** Chromogenic zymography activated with phenol red glycerol, tributyrates and olive oil as substrates. **C.** Transfer zymography activated with glycerol tributyrates and olive oil as substrates.

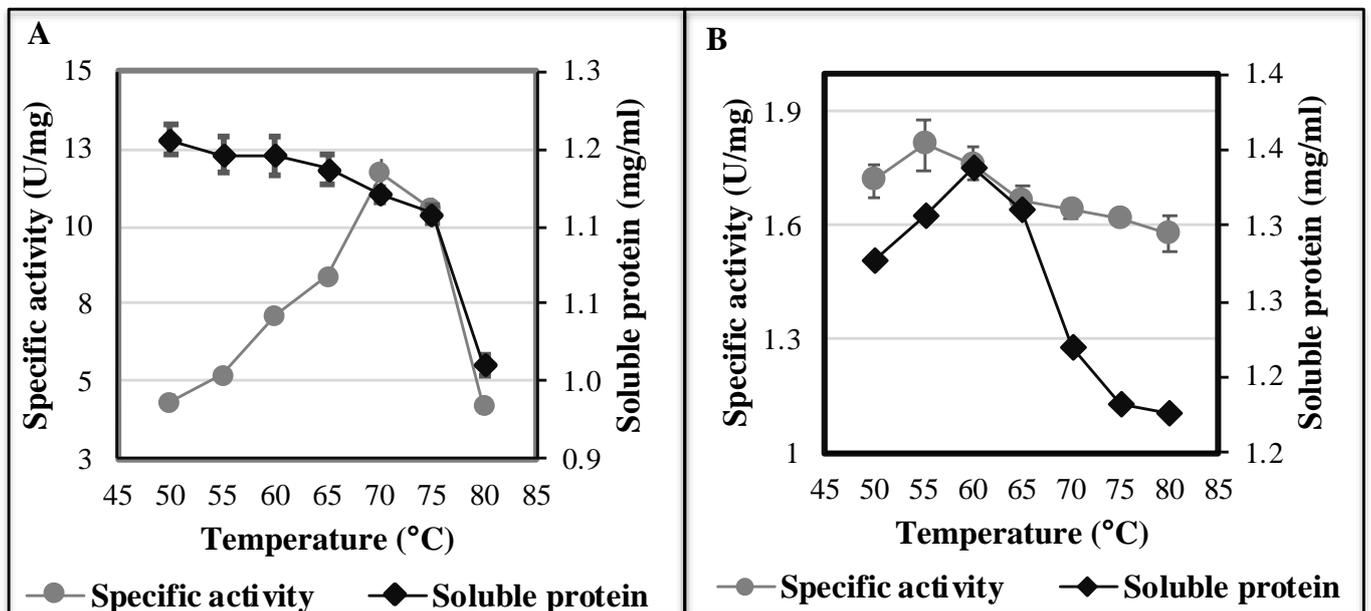


Figure 3. Lipase activity and soluble protein from cell lysates of *Geobacillus stearothermophilus* (**A**) and *Anoxybacillus flavithermus* (**B**) in supernatants after heat treatment. Error bars represent the standard deviation.

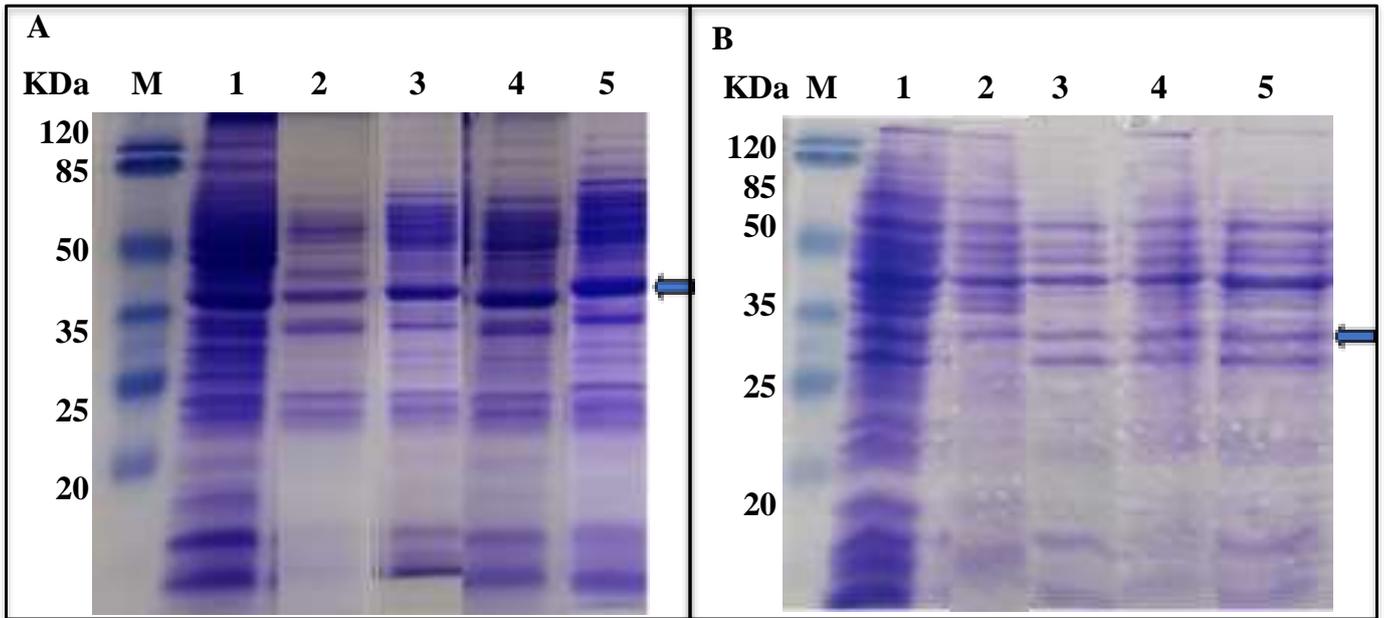


Figure 4. SDS-PAGE analysis of lipase isolates from *Geobacillus stearothermophilus* (A) and *Anoxybacillus flavithermus* (B), both gels stained with Coomassie blue. M is molecular weight markers. Lanes 1A & B. Crude lysate, 2A. Heat treatment at 70°C supernatant, 2B. Heat treatment at 55°C supernatant, 3A. Ethanol precipitate at 80%, 3B. Ethanol precipitate at 60%, 4A. Ammonium sulfate precipitate at 95%, 4B. Ammonium sulfate precipitate at 80%, 5A & B. Gel filtration fraction. Arrows indicate the lipase bands from *Geobacillus stearothermophilus* and *Anoxybacillus flavithermus* samples with molecular weights of 42 kDa and 33 kDa respectively.

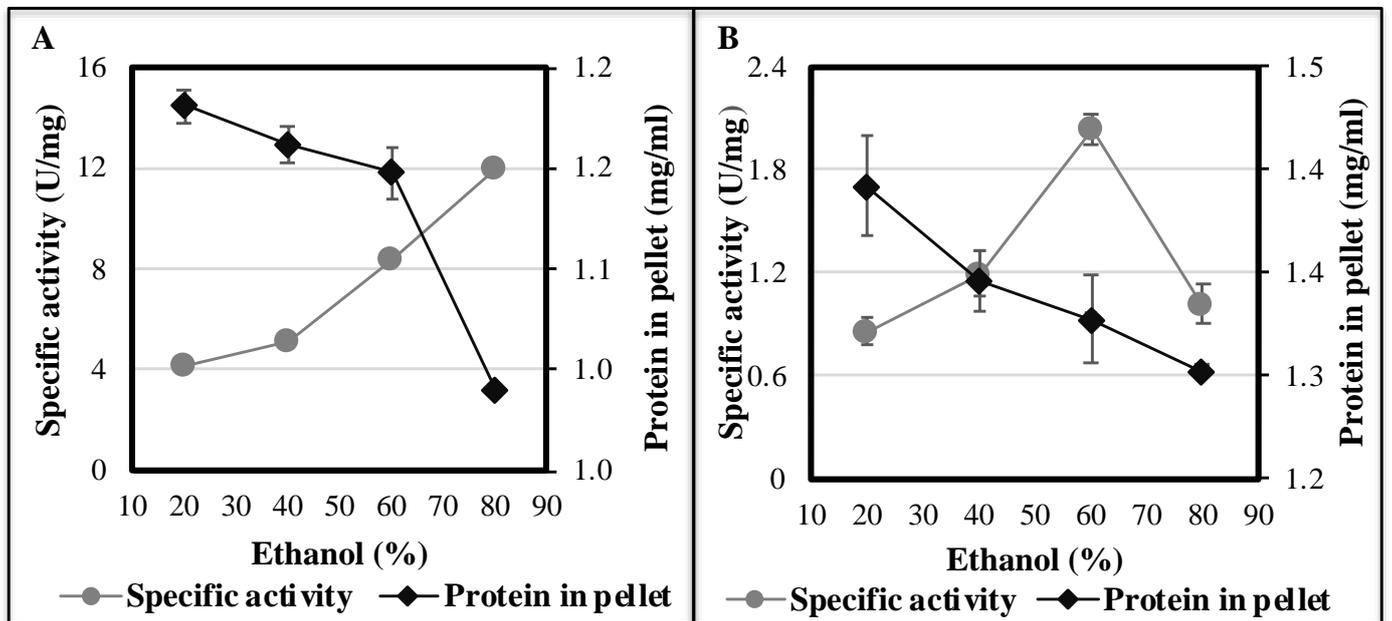


Figure 5. Lipase activity and protein concentration in pellets of cell lysates from *Geobacillus stearothermophilus* (A) and *Anoxybacillus flavithermus* (B) after ethanol precipitation. Error bars represent the standard deviation.

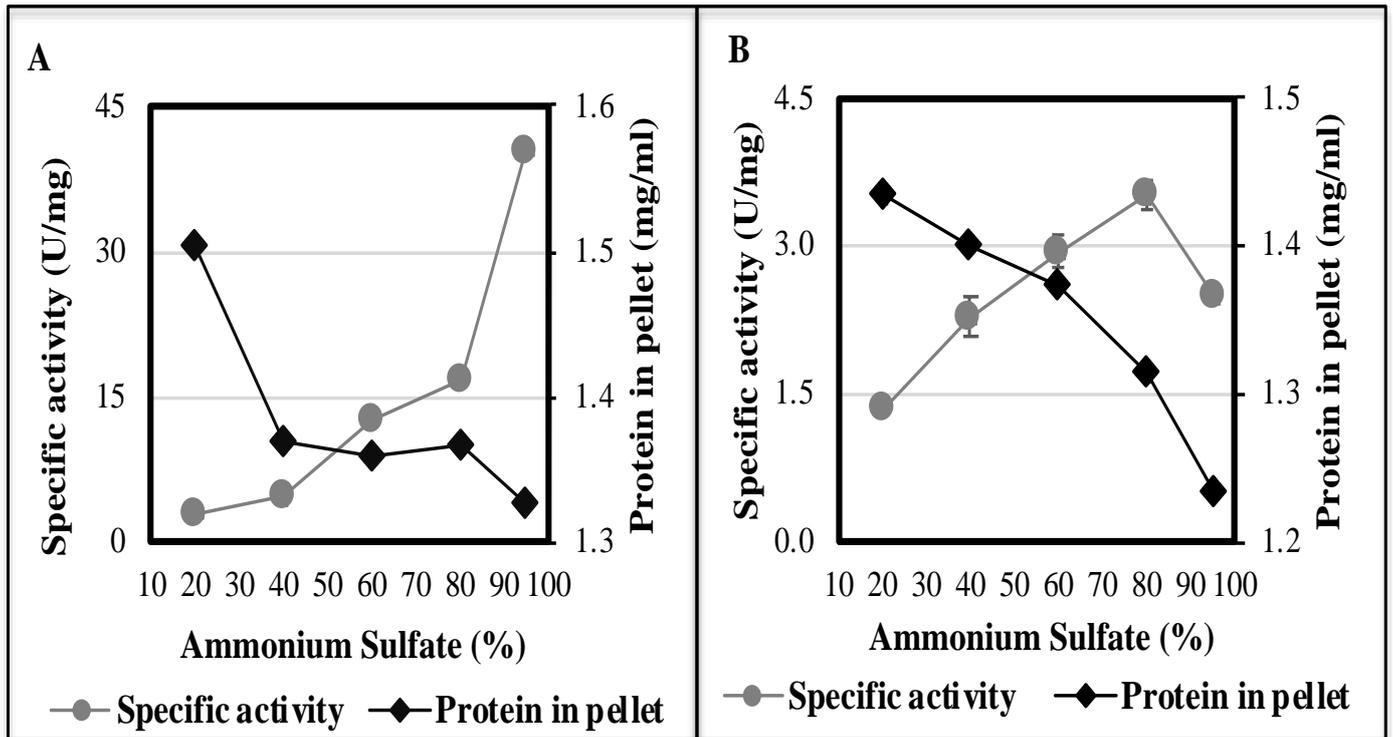


Figure 6. Lipase activity and protein concentration in pellets after dialysis of cell lysates from *Geobacillus stearothermophilus* (A) and *Anoxybacillus flavithermus* (B) after ammonium sulfate precipitation. Error bars represent the standard deviation.

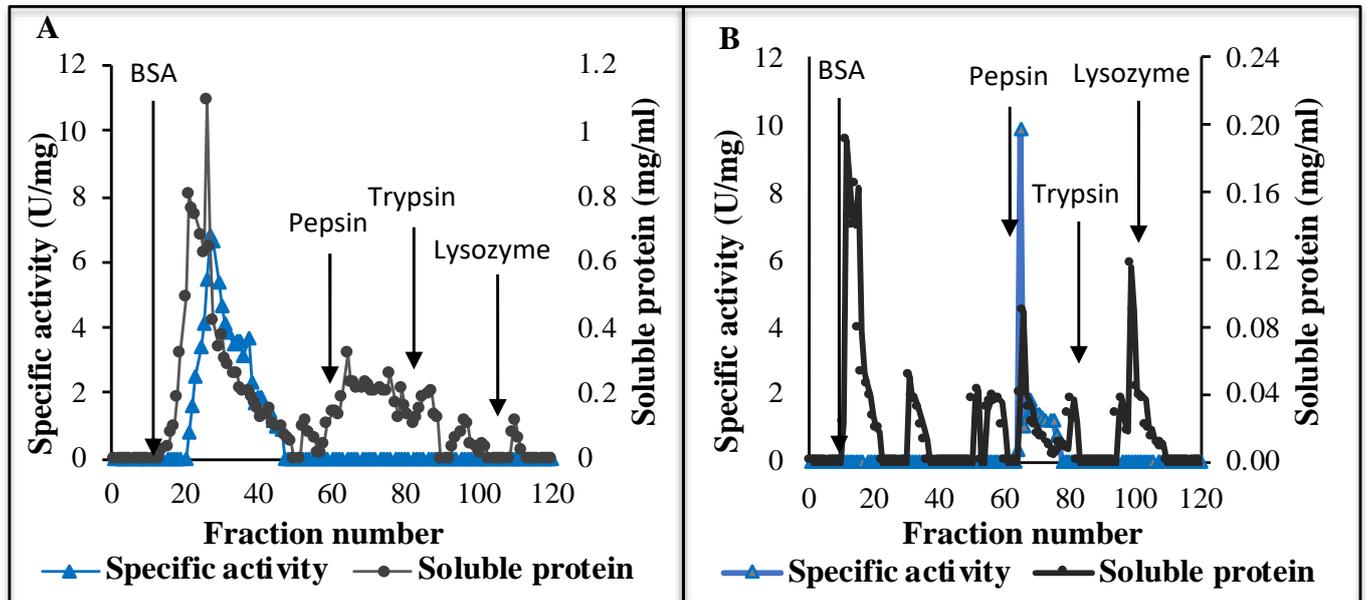


Figure 7. Size-exclusion chromatography on Sephadex G-100. Specific lipase activity and soluble protein from cell lysates from *Geobacillus stearothermophilus* (A) and *Anoxybacillus flavithermus* (B) were loaded on a Sephadex G-100 column (36.0 cm x 2 cm) equilibrated in Tris HCL buffer (pH 7.5). The flow rate was 0.5 ml/min, and the chromatography was carried out at 4°C. Molecular weight protein standards were used to calibrate the column and calculate the lipases molecular weights based on retention times.

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